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Outline

X-ray absorption spectroscopy (XAS)

- Absorption process
- Total electron yield mode
- Examples

X-ray Magnetic Circular Dichroism (XMCD)

- Basics
- Example: Magnetocrystalline Anisotropy

Closer look at the absorption process

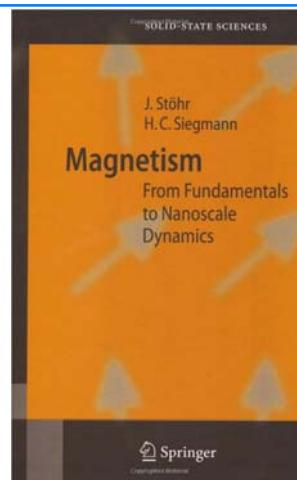
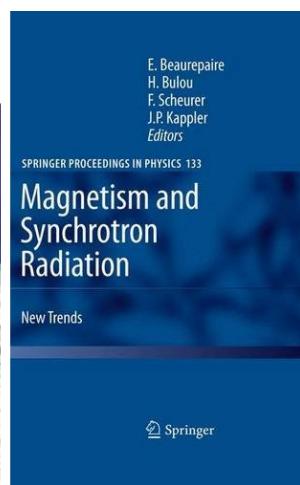
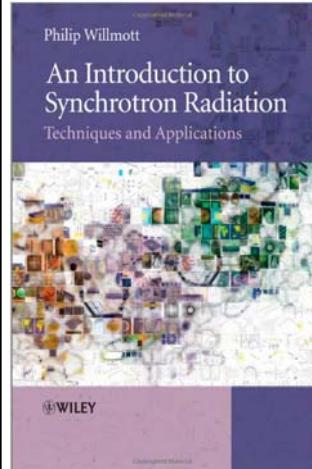
- Multiplet effects
- Example: Interface effect in Exchange Bias system

X-ray Magnetic Linear Dichroism (XMLD)

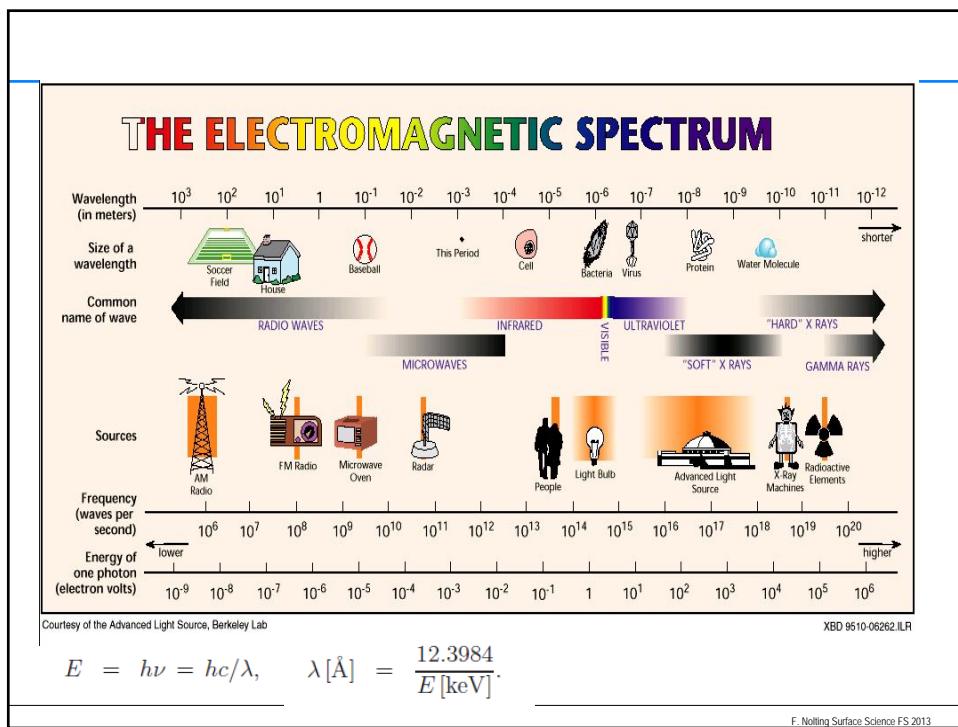
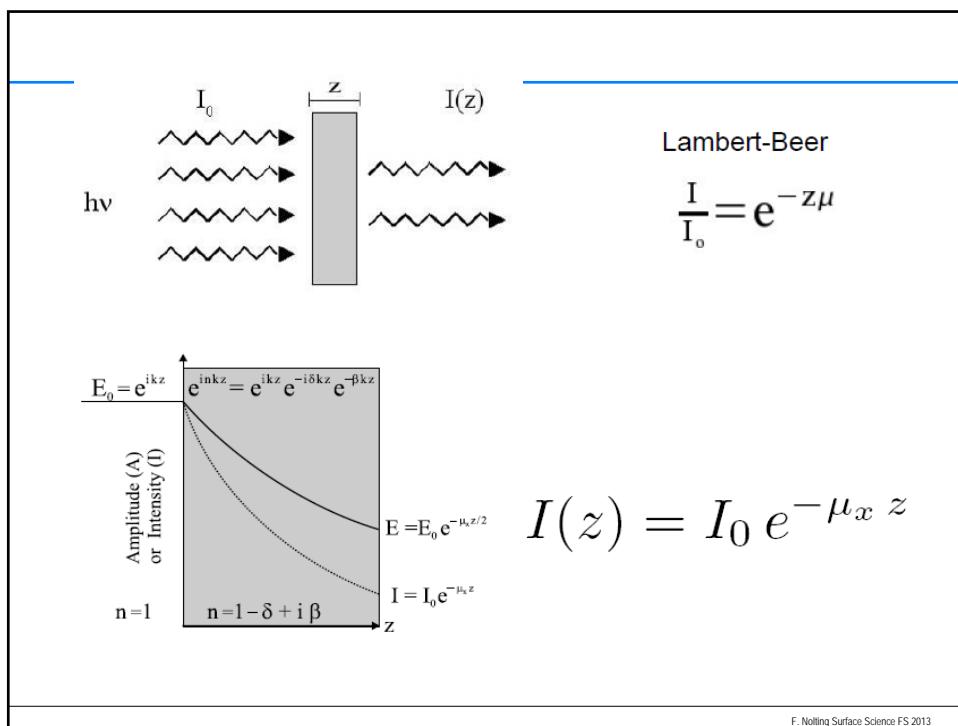
- Basics

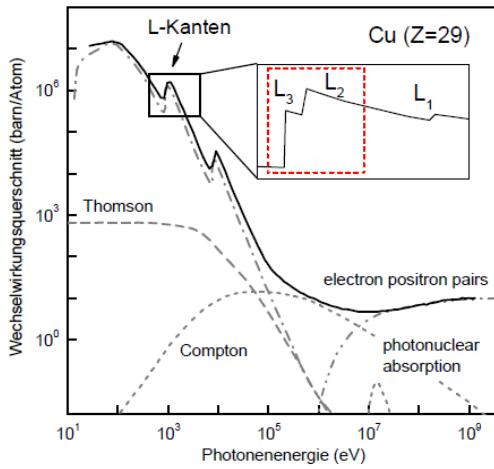
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Some good books



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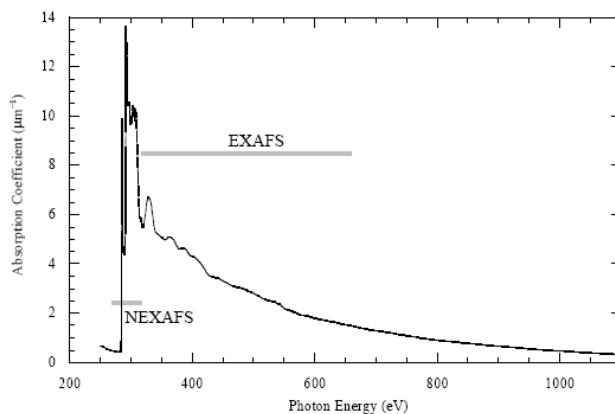




J. H. Hubbel, H. A. Gimm und I. Øverbø, *Pair, triplet, and total atomic cross sections (and mass attenuation coefficients) for 1 MeV-100 GeV photons in elements Z = 1 to 100*, J. Phys. Chem. Ref. Data 9, 1023 (1980).

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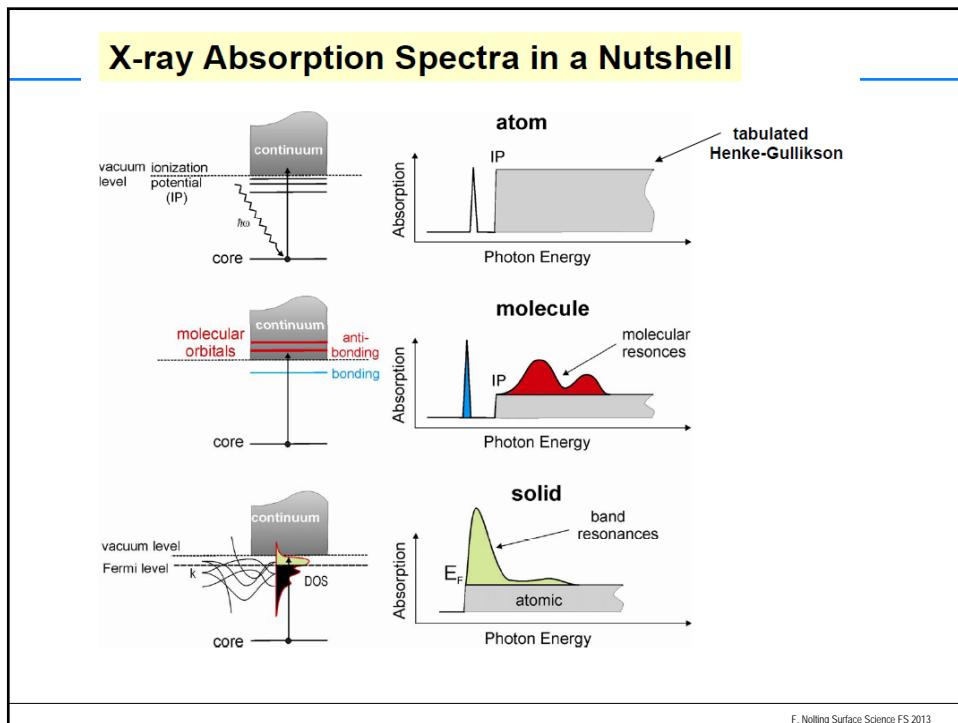
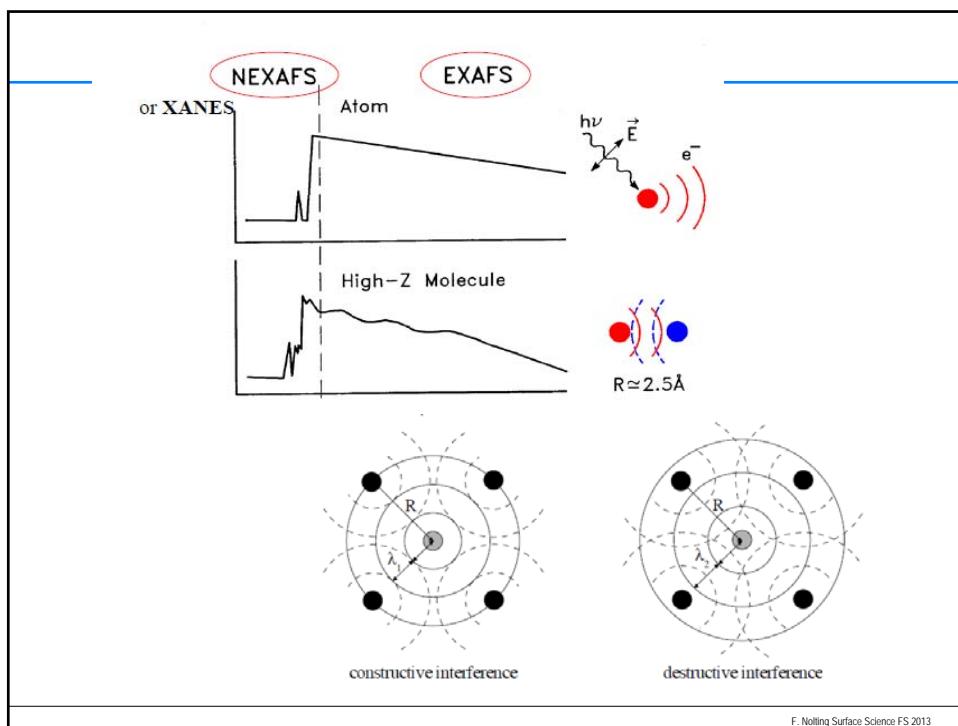
Absorption of Photons in the Soft X-ray Range



Near Edge X-ray Absorption Fine Structure
reflects density of unoccupied states
Also called XANES

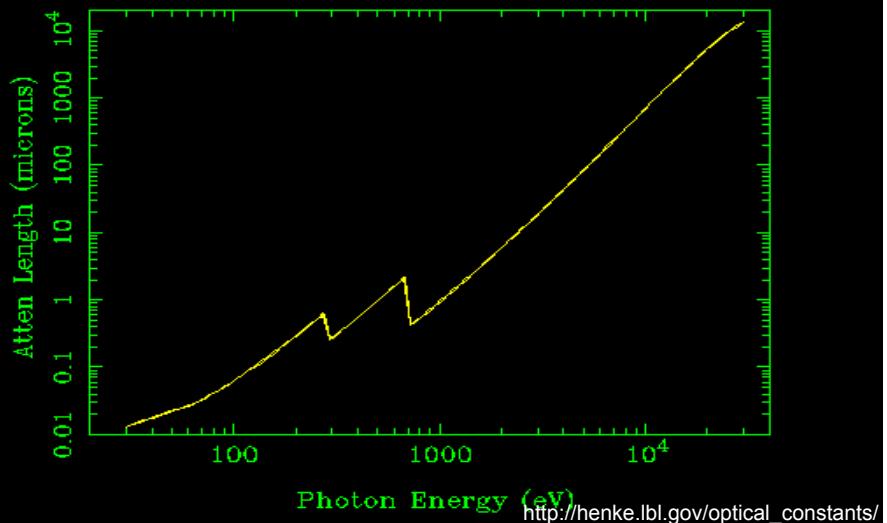
Extended X-ray Absorption Fine Structure
reflects spatial location of neighboring atoms

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Absorption - Teflon

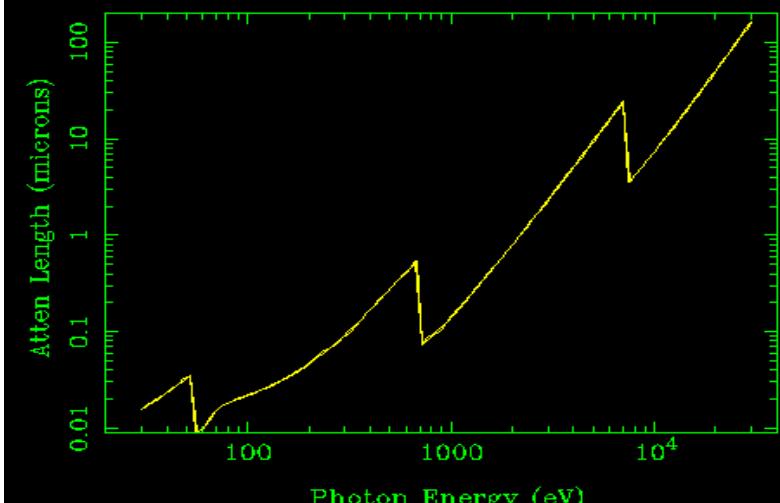
C2F₄ Density=2.2, Angle=90.deg



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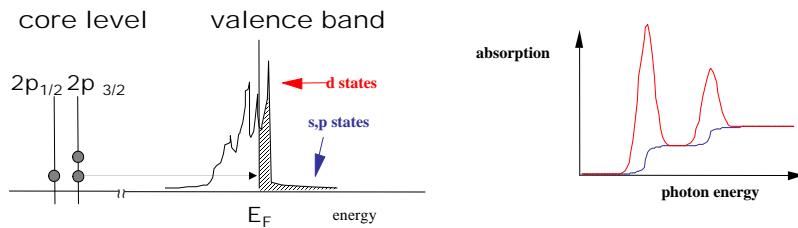
Absorption - Iron

Fe Density=7.874, Angle=90.deg



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Interaction of electromagnetic wave with charge



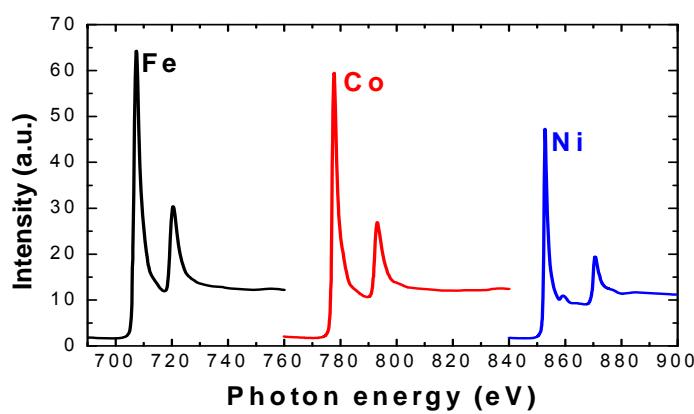
$$\text{Absorption} \sim \frac{\text{Transition matrix}}{\text{Final state} \quad \text{Initial state}} \cdot \text{Density of final states}$$

Fermi's golden rule
in dipole approximation
wavelength is large compared to charge
1000 eV corresponds to 1.2 nm
2p core radius is about 0.01 nm

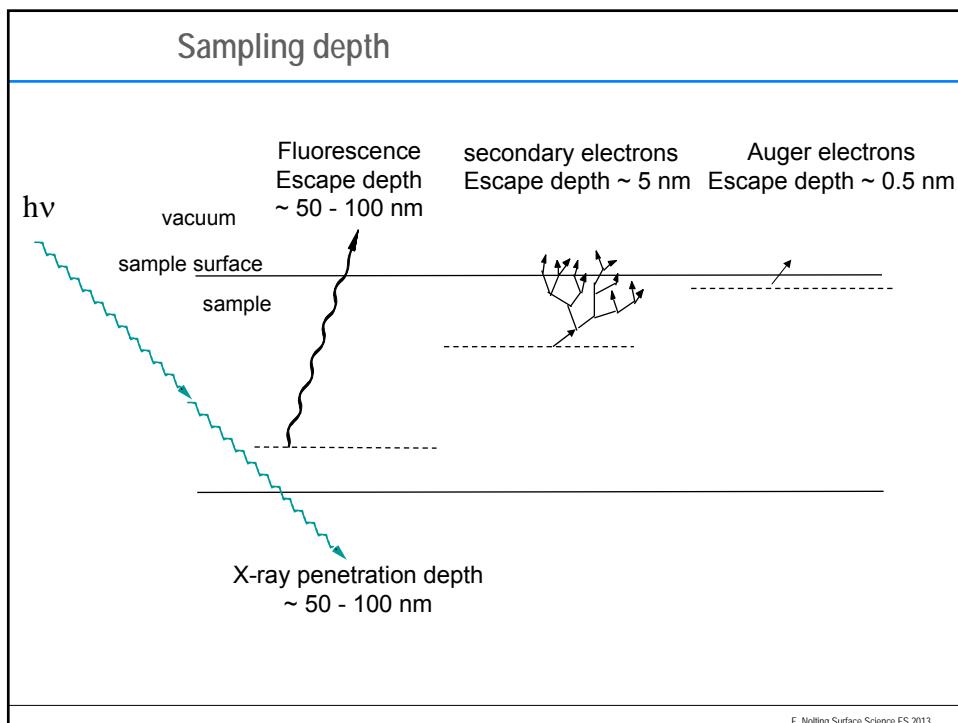
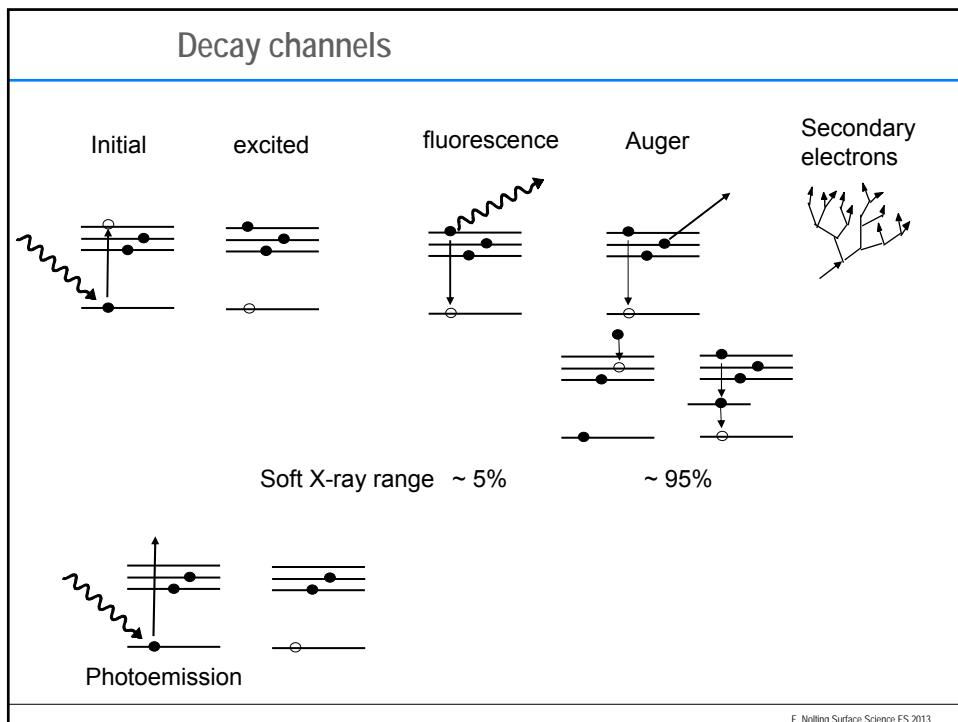
$$\mu \propto |\langle f | e \cdot p | i \rangle|^2 \rho(E)$$

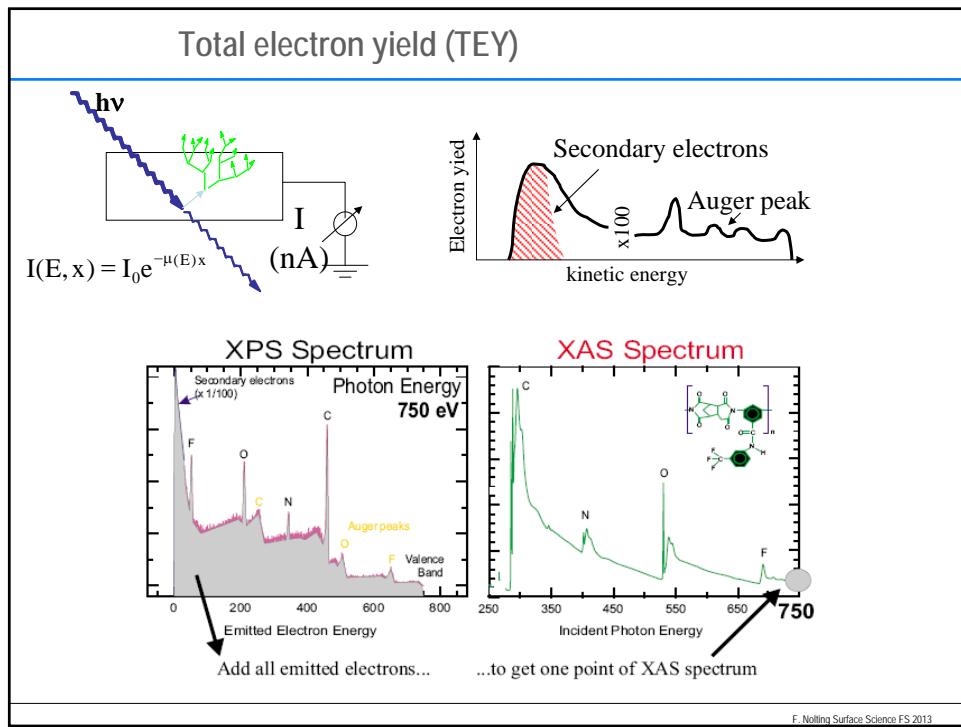
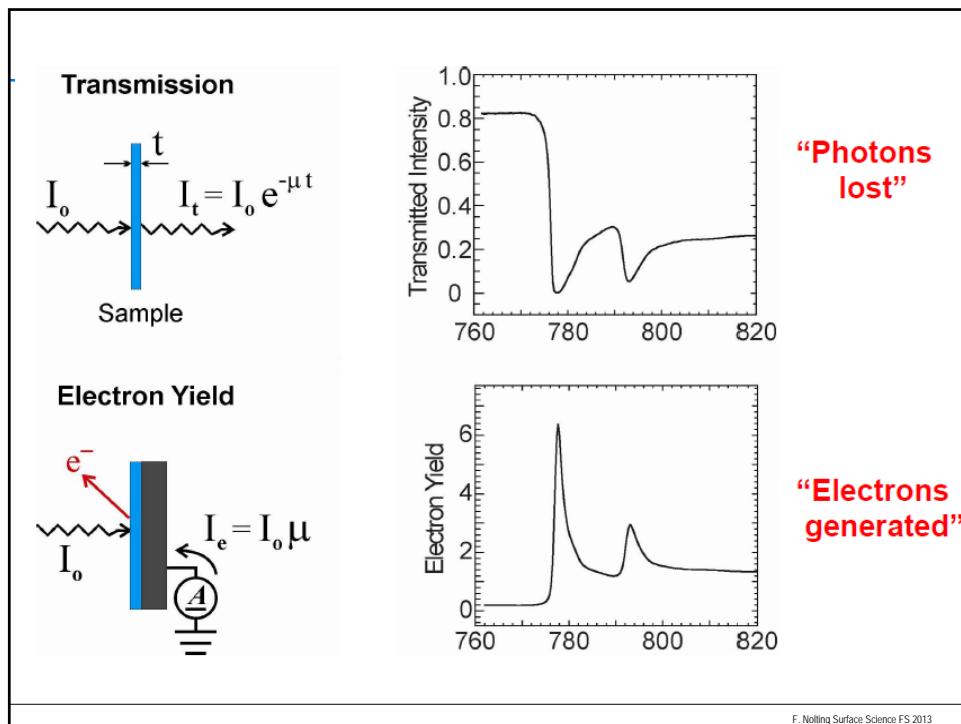
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Element specific

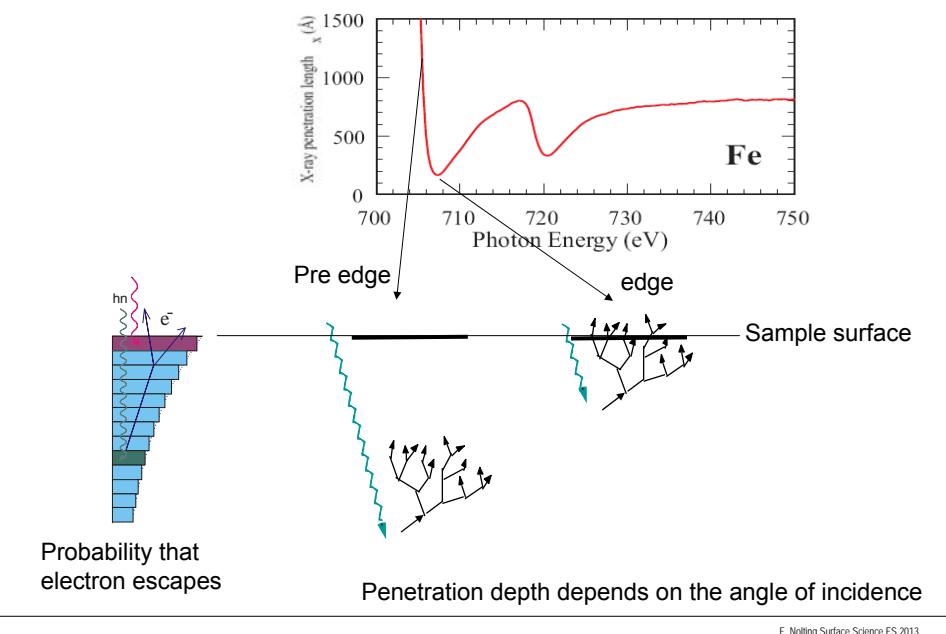


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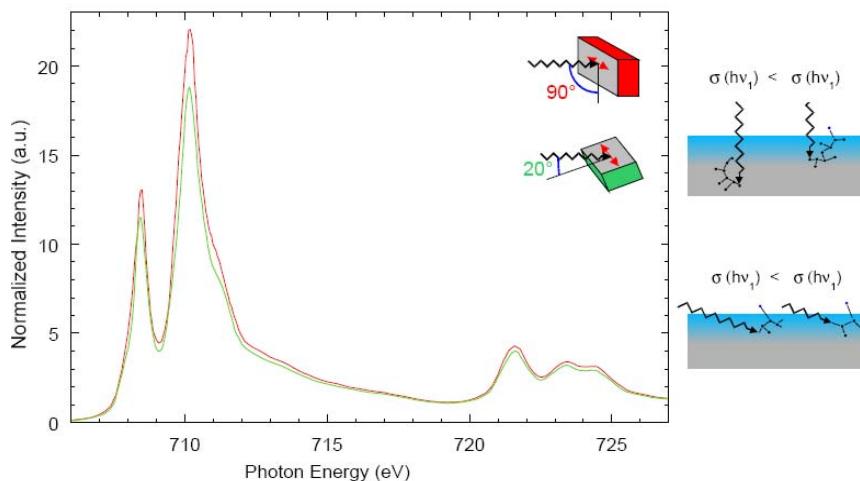


Why is TEY proportional to absorption coefficient



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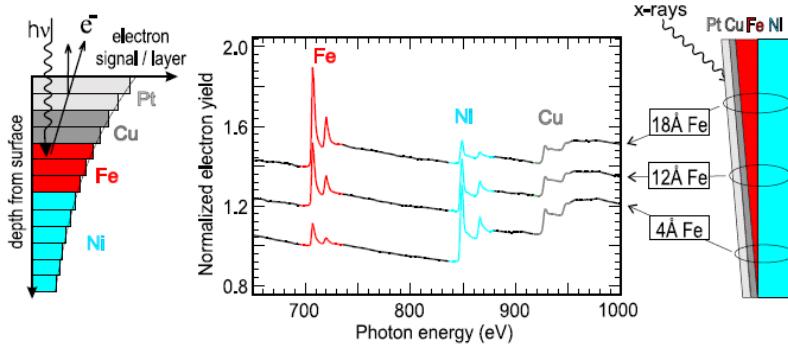
Saturation Effects in TEY Detection



J. Lüning et al, PRB 67, 214433 (2003)

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"Dismantle" a Multilayer



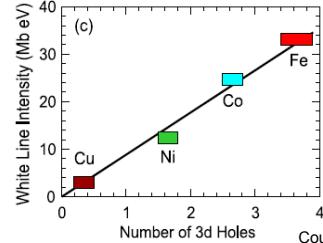
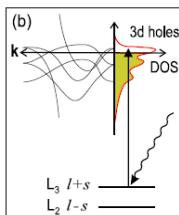
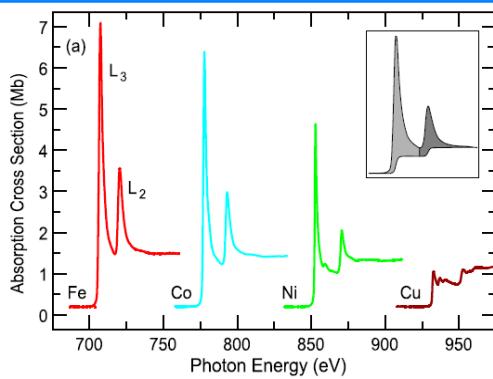
The probability that an electron generated at depth t is $\sim e^{-t/\lambda}$
With λ the electron escape length typically a few nm

Courtesy J. Stöhr
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Determine number of 3d holes

Charge sum rule

Integrated intensity
is proportional to
number of empty
valence states



Courtesy J. Stöhr
F. Nolting Surface Science FS 2013

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- Crash class nanomagnetism probed with X-rays
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- Example: Interface effect in Exchange Bias system

X-ray Magnetic Linear Dichroism (XMLD)

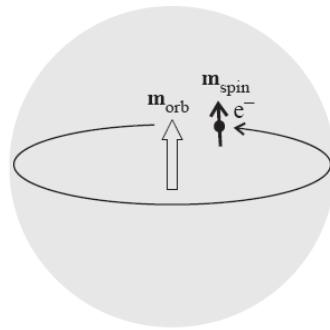
- Basics

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Crash class nanomagnetism probed with X-rays

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Source of magnetism (atomic)



Spin moment $\sim 1.5 \mu_B / \text{atom}$ isotropic

Orbital moment $\sim 0.1 \mu_B / \text{atom}$ isotropic/anisotropic

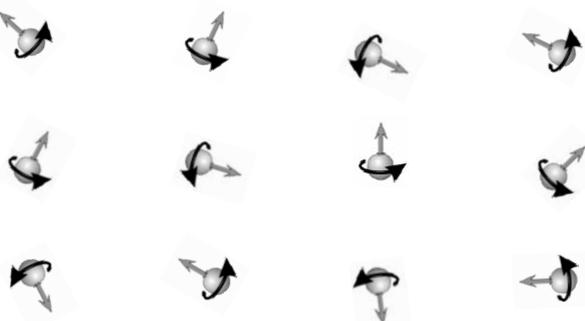
They interact via the spin-orbit coupling $L \cdot S$

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Switching on the interaction

Atoms have an magnetic moment

Without interaction (and no applied magnetic field) they point in random directions and no macroscopic magnetic field is created

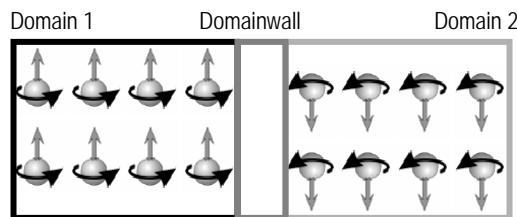


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Switching on the interaction

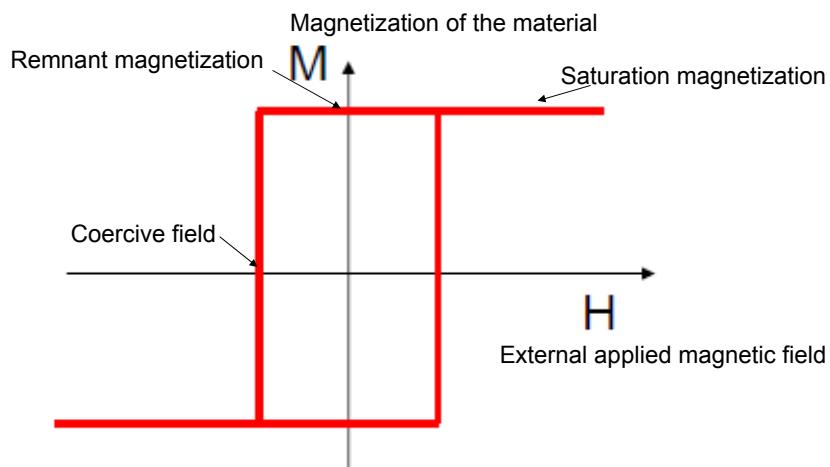
Atoms have an magnetic moment

With interaction they can align to each other and can create macroscopic magnetic field



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Hysteresis loop

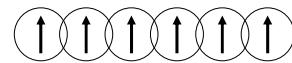


Responses of a material to an applied magnetic field is described by the hysteresis loop

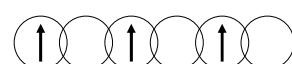
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Different magnetic interactions

Direct exchange
Overlapping wavefunctions



superexchange
Overlapping wavefunctions via
non-magnetic atoms

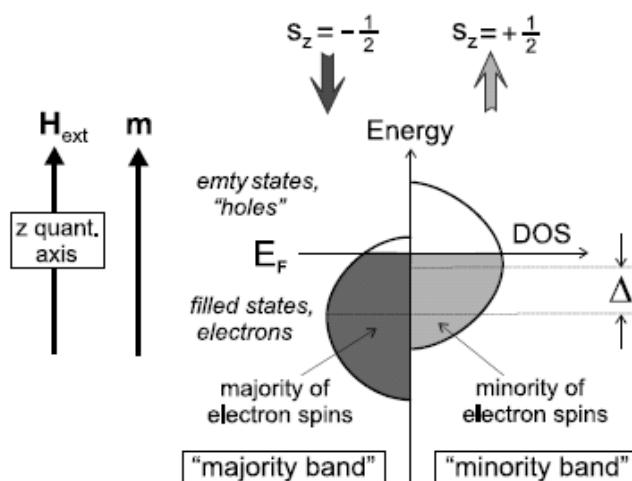


Indirect exchange
Exchange via delocalised electrons



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Stoner Model for 3d Band and Nomenclature



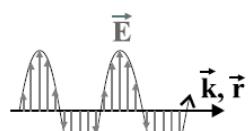
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Polarized Photons

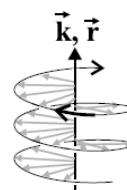
$$\mathbf{E} = (\mathbf{B} \times \mathbf{k}_0) c$$

$$\mathbf{B} = (\mathbf{k}_0 \times \mathbf{E}) / c$$

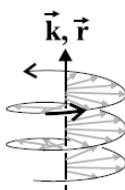
Linear polarization



Left circular polarization
space



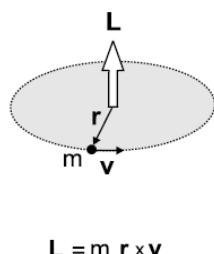
Right circular polarization



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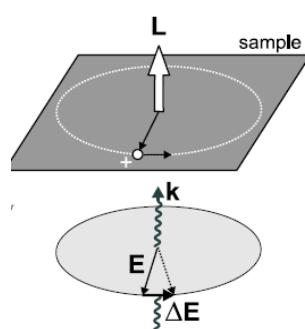
Polarized Photons

Angular momentum of orbiting mass



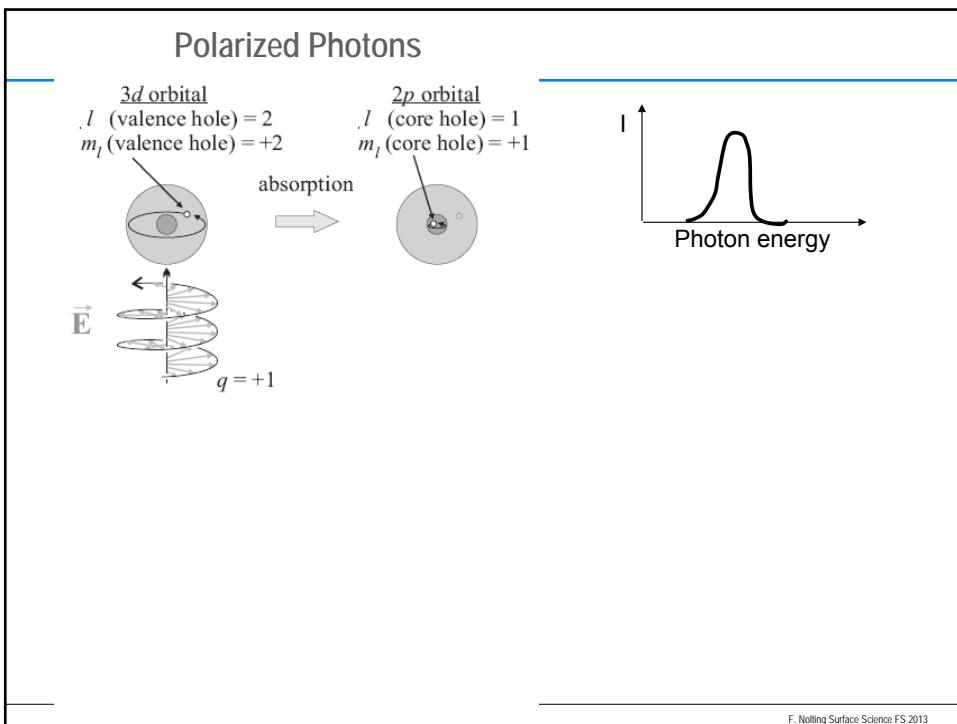
$$\mathbf{L} = m \mathbf{r} \times \mathbf{v}$$

Photon angular momentum

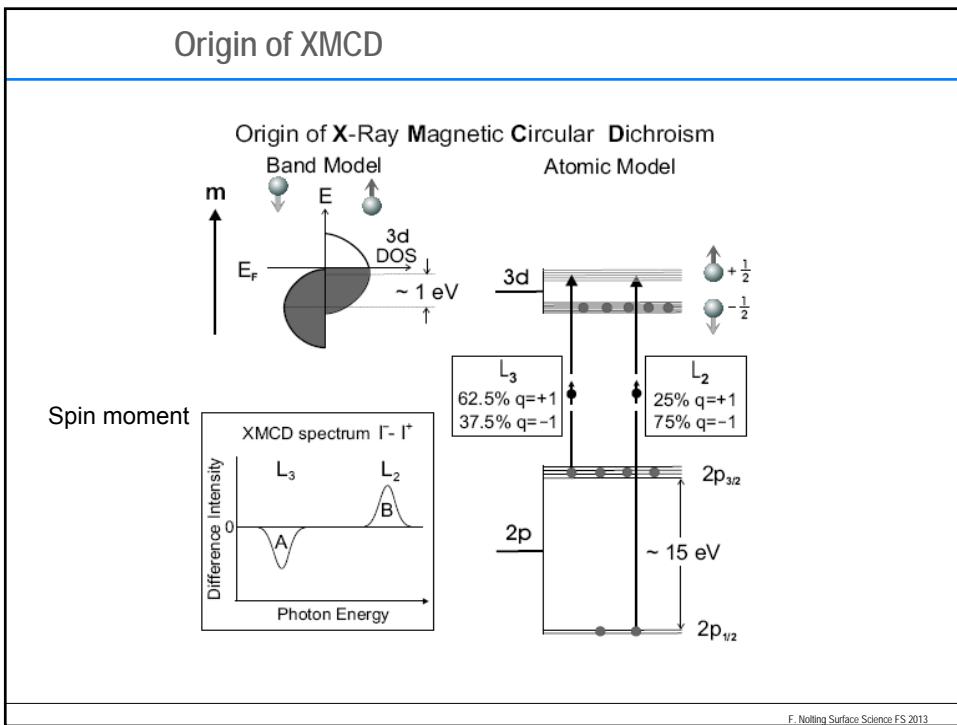


Angular momentum conservation

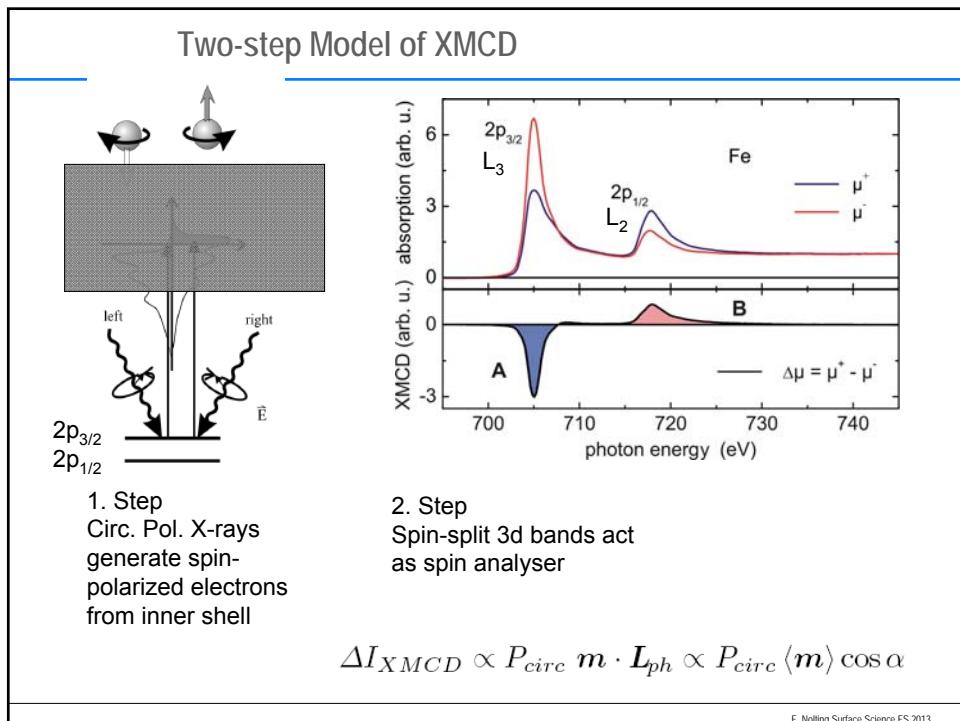
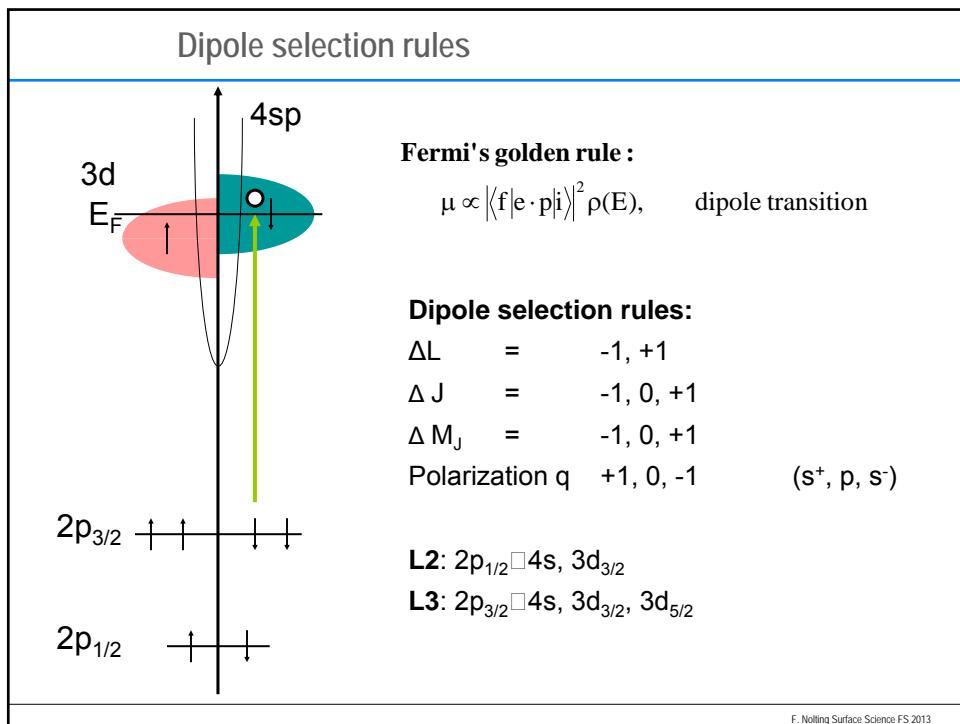
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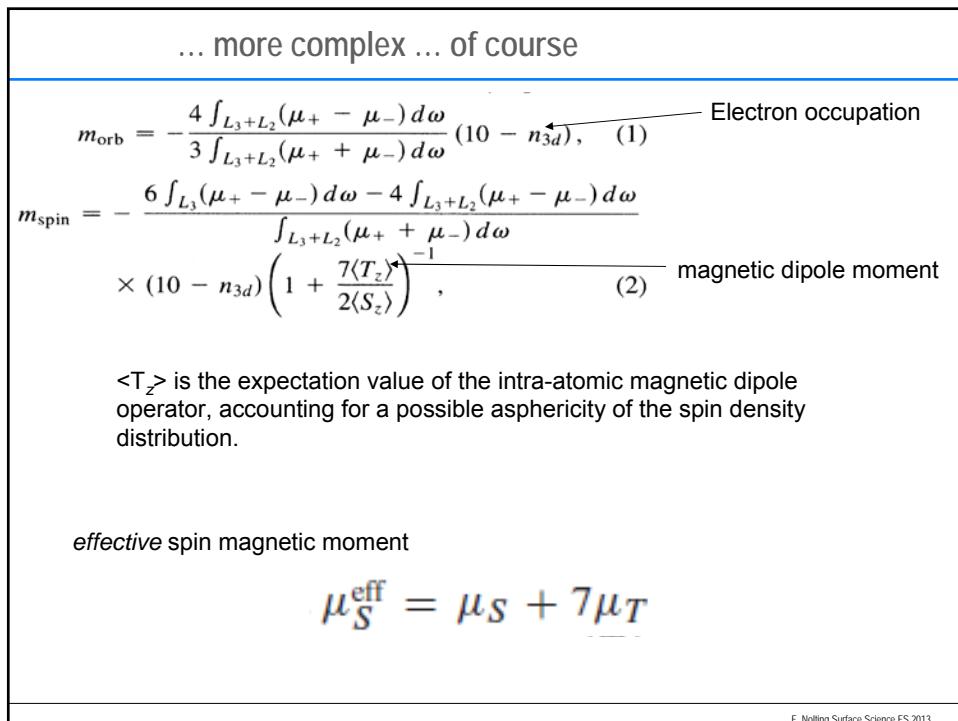
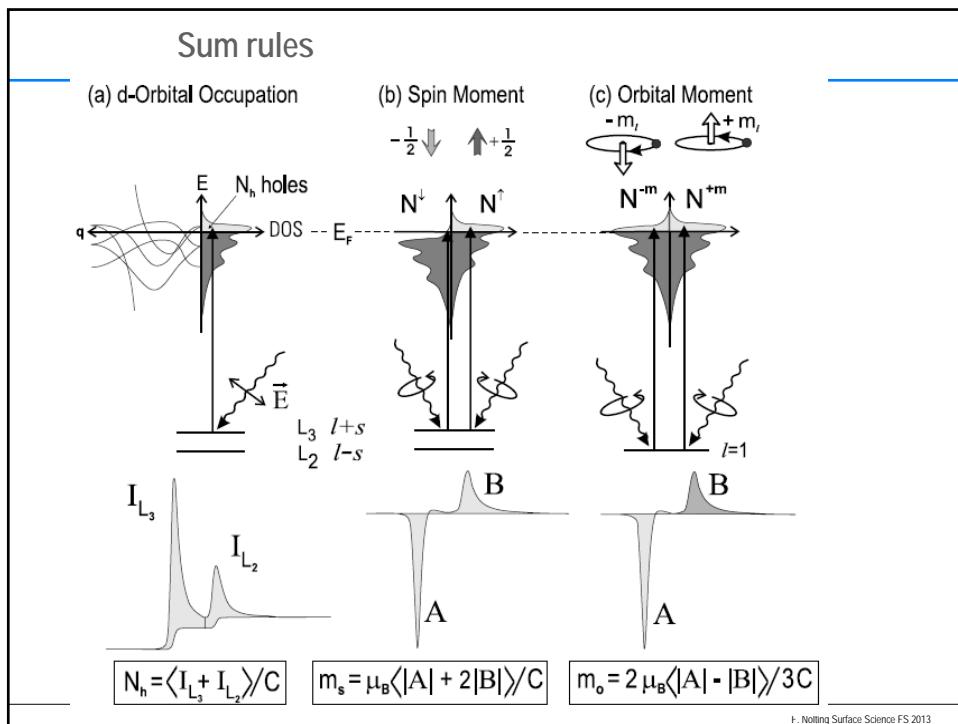


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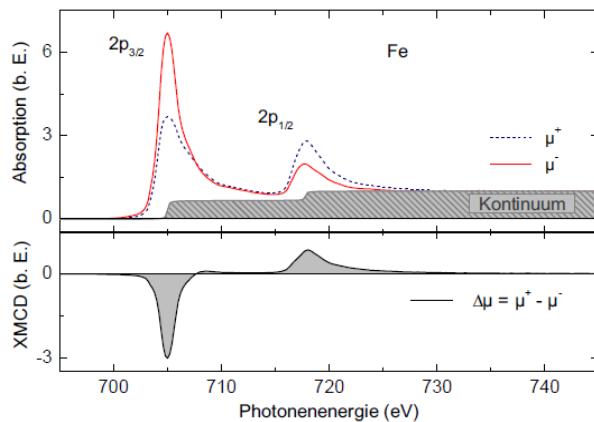


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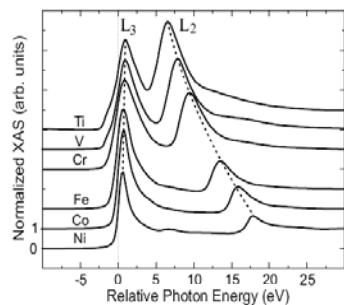
Only contribution to the resonance absorption should be considered



Taken from PhD Thesis Armin Kleibert, 2005

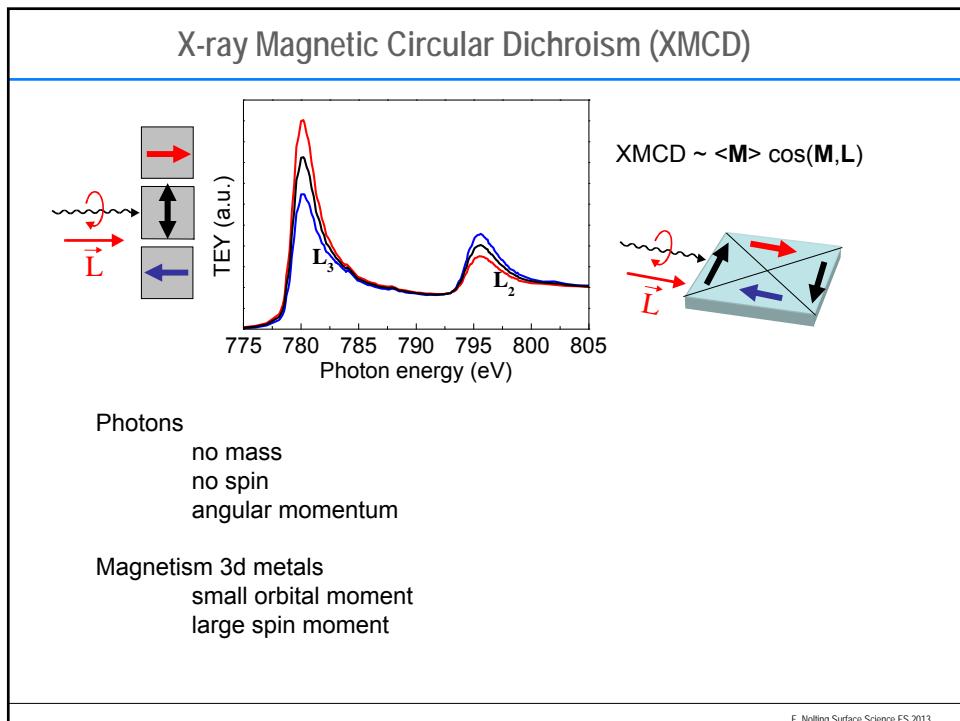
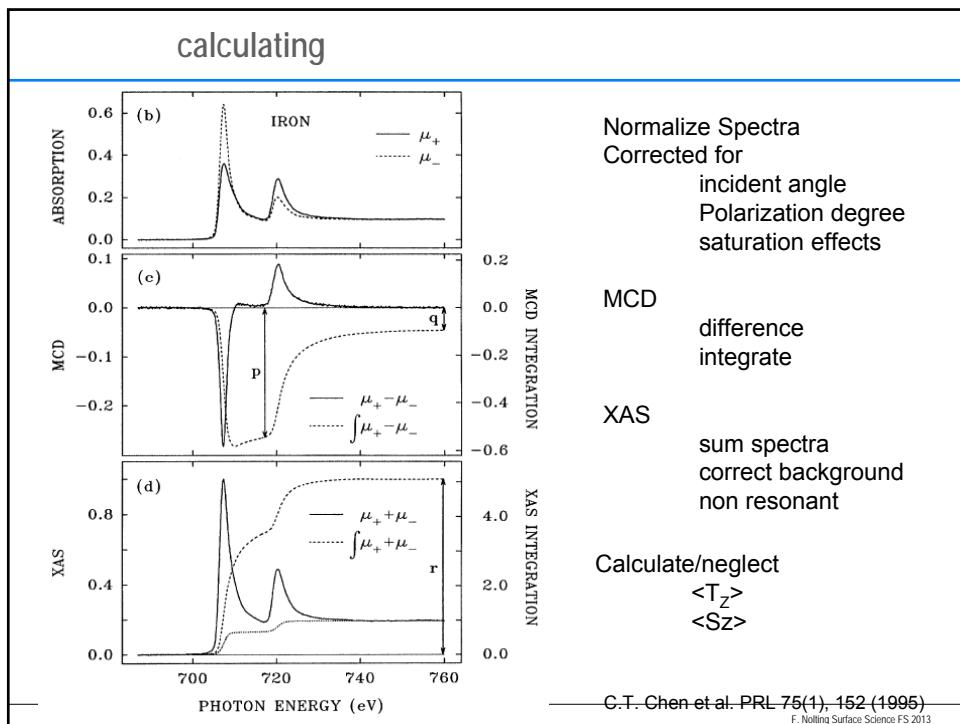
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... more complex ... of course



L_3 and L_2 must be separated

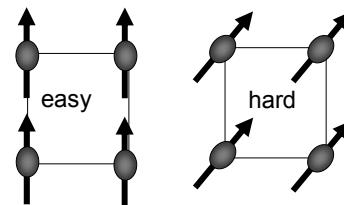
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XMCD in action

Study Magneto-crystalline anisotropy

Magnetic Anisotropy
preferential magnetization along axes
easy / hard axis
(magneto-crystalline anisotropy)



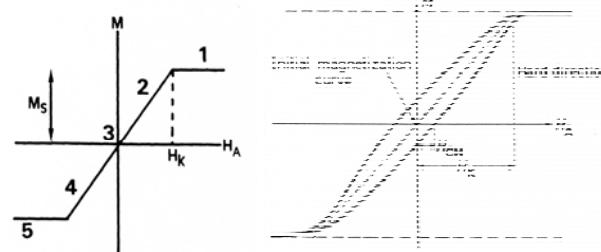
The magneto-crystalline anisotropy is the energy that it takes to rotate the magnetization from the “easy” direction into the “hard” direction

J. Stöhr, JMMM 200 (1999) 470 – 497
Reiko Nakajima PhD Thesis 1998

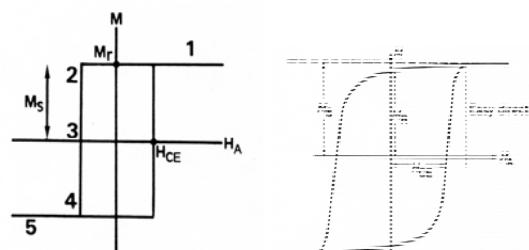
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Hysteresis loop – anisotropy

hard axis

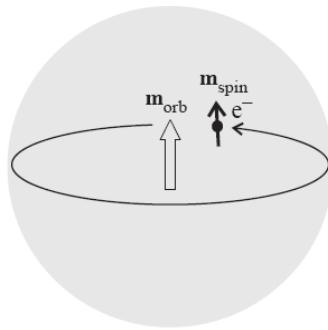


easy axis



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Magneto-crystalline anisotropy



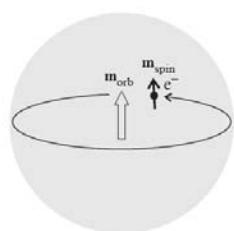
Spin moment $\sim 1.5 \mu_B / \text{atom}$ isotropic

Orbital moment $\sim 0.1 \mu_B / \text{atom}$ isotropic/anisotropic

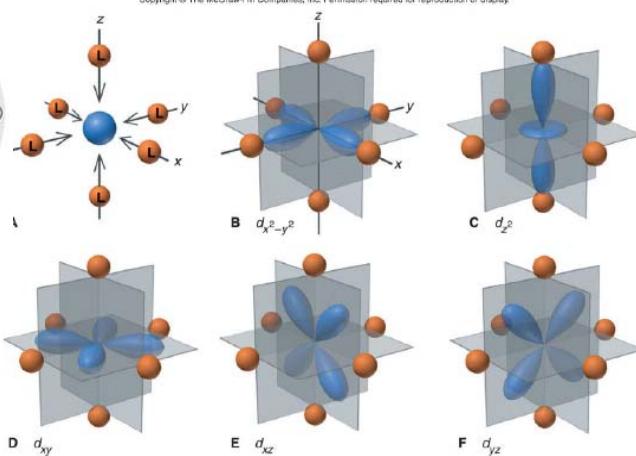
They interact via the spin-orbit coupling $L \cdot S$

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Anisotropy



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Bruno model

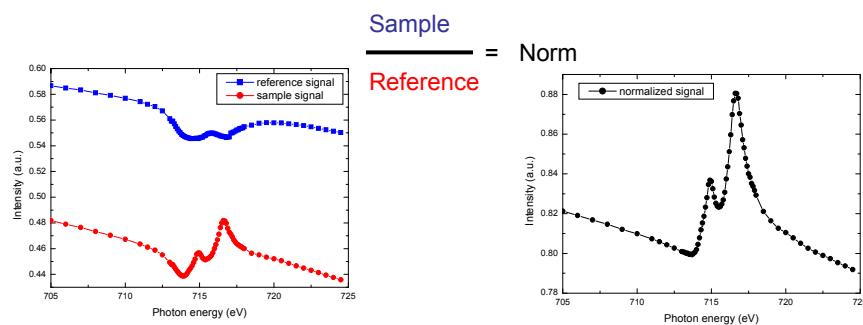
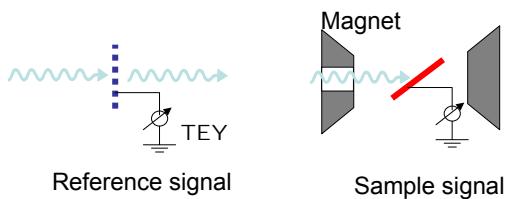
The Bruno model states that the orbital moment is larger along the easy magnetization direction, and that the difference between the orbital moments along the easy and hard directions is proportional to the magneto-crystalline anisotropy

$$\Delta E_{so} = \zeta [\langle \mathbf{L} \cdot \mathbf{S} \rangle_{hard} - \langle \mathbf{L} \cdot \mathbf{S} \rangle_{easy}] = \frac{\zeta}{4\mu_B} (m_o^{easy} - m_o^{hard}) > 0$$

P. Bruno, PRB 39, 865 (1989)

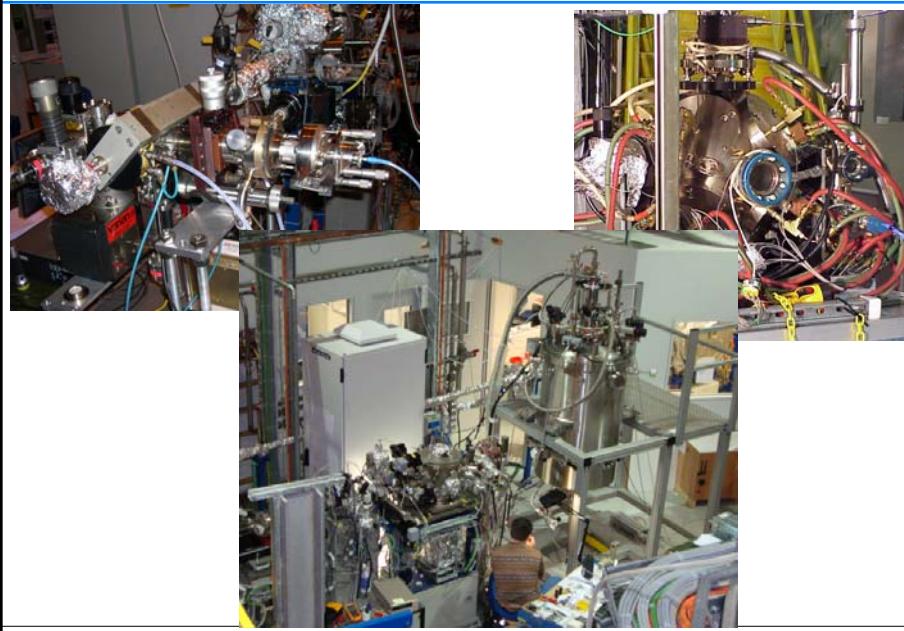
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How do we measure



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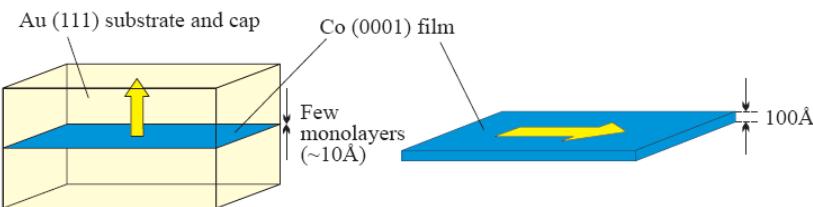
How do we measure



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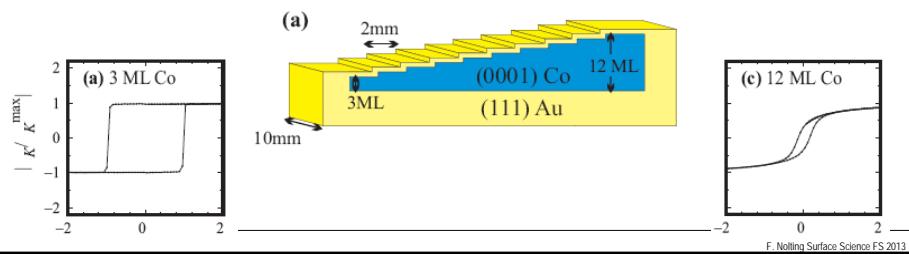
Test system

Magnetic anisotropy in Co (0001) films

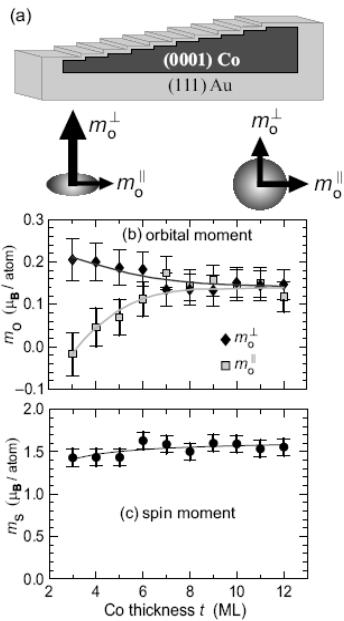


Perpendicular magnetic anisotropy
(preferred magnetization out-of-plane)

Thin film shape anisotropy
(preferred magnetization in-plane)



Results



Thin film

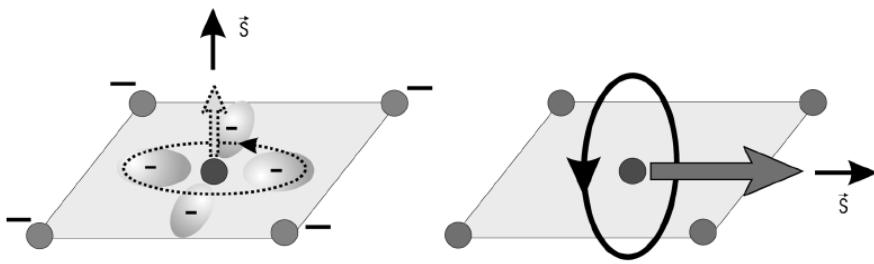
Orbital moment is anisotropic and larger out-of-plane

Thick film

Orbital moment is isotropic
shape anisotropy is dominating

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Simple picture – Ligand fields



Free monolayer

in-plane moment

Multilayer with stronger out-of-plane bonding

out-of-plane moment

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Ferromagnetism in one-dimensional monatomic metal chains

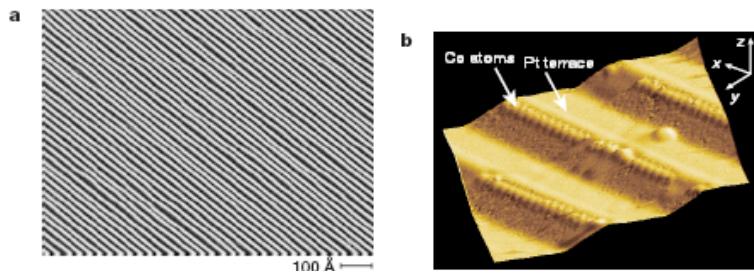
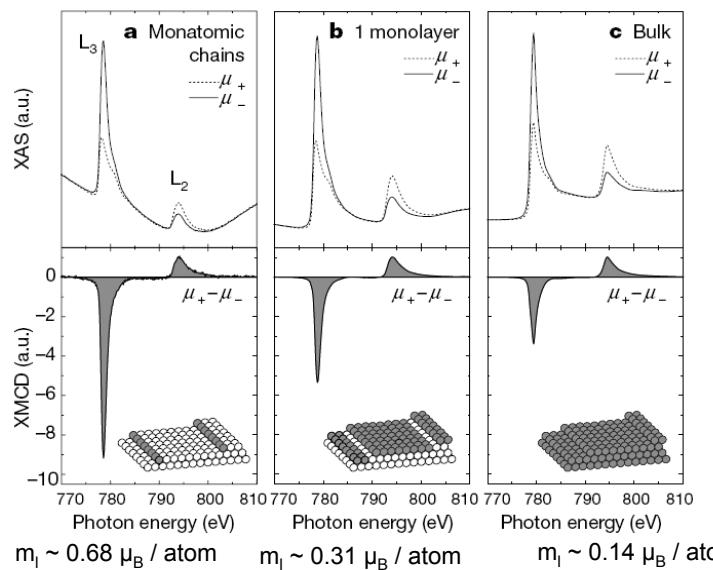


Figure 1 STM topographs of the Pt(997) surface. **a**, Periodic step structure (each white line represents a single step). The surface has a 6.45° miscut angle relative to the (111) direction; repulsive step interactions result in a narrow terrace width distribution centred at 20.2 Å with 2.9 Å standard deviation. **b**, Co monatomic chains decorating the Pt step edges (the vertical dimension is enhanced for better contrast). The monatomic chains are obtained by evaporating 0.13 monolayers of Co onto the substrate held at $T = 260\text{ K}$ and previously cleaned by ion sputtering and annealing cycles in ultrahigh vacuum (UHV). The chains are linearly aligned and have a spacing equal to the terrace width.

P. Gambardella et al. Nature 416, 301 (2002)

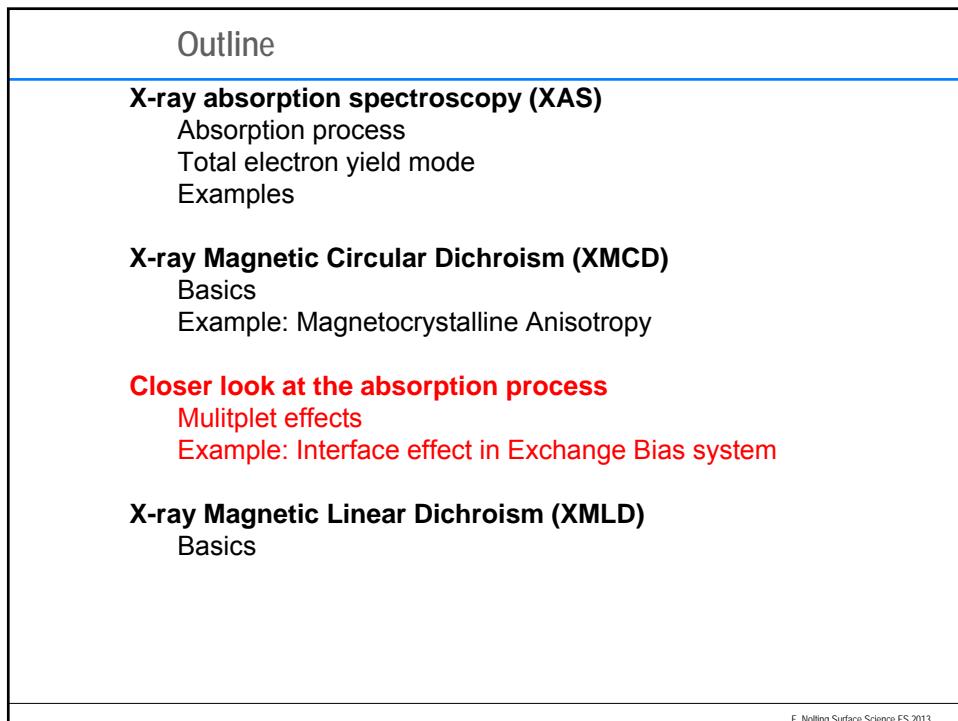
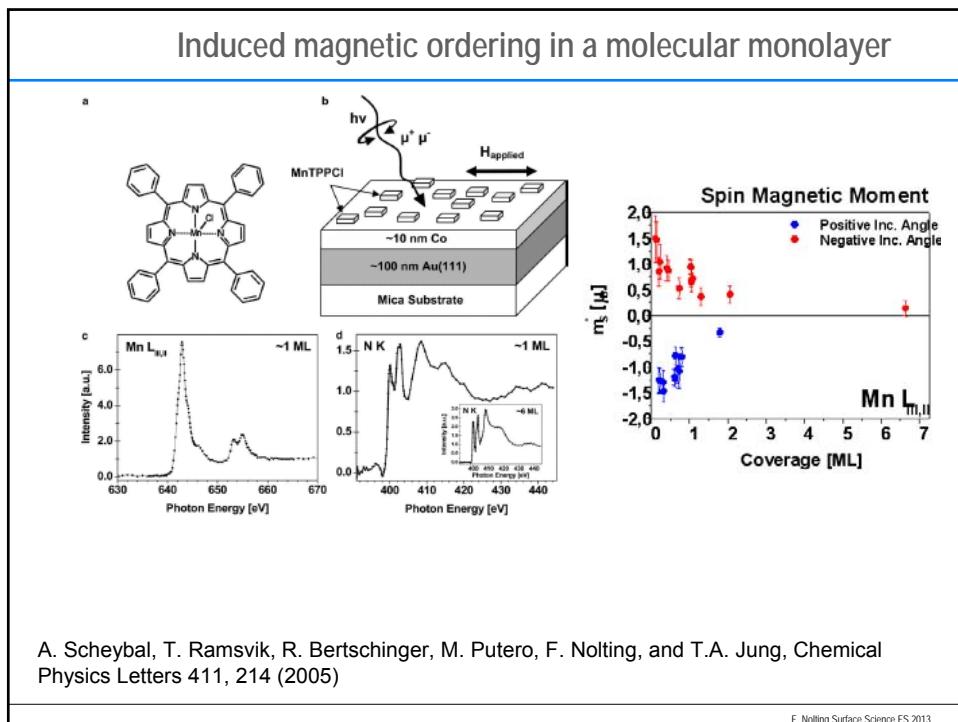
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Increased orbital moment in Co chains



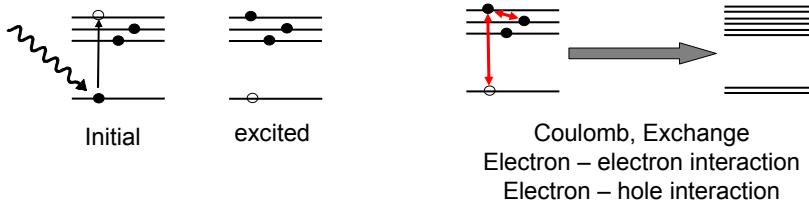
P. Gambardella et al. Nature 416, 301 (2002)

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Interactions

Single electron



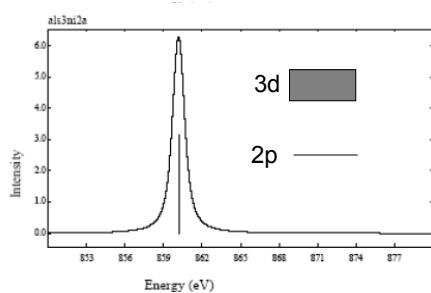
Valence – Valence interaction : many body effects

Valence – Core interaction : multiplet effects

Hybridization between ground state and final state
leads to a multiplet structure of the spectrum

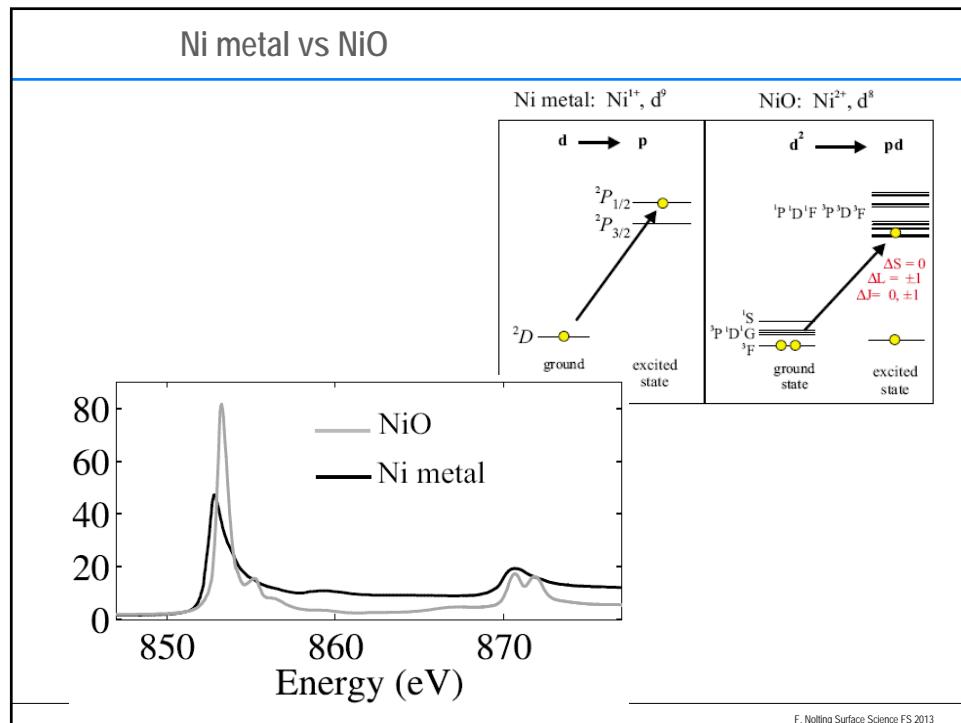
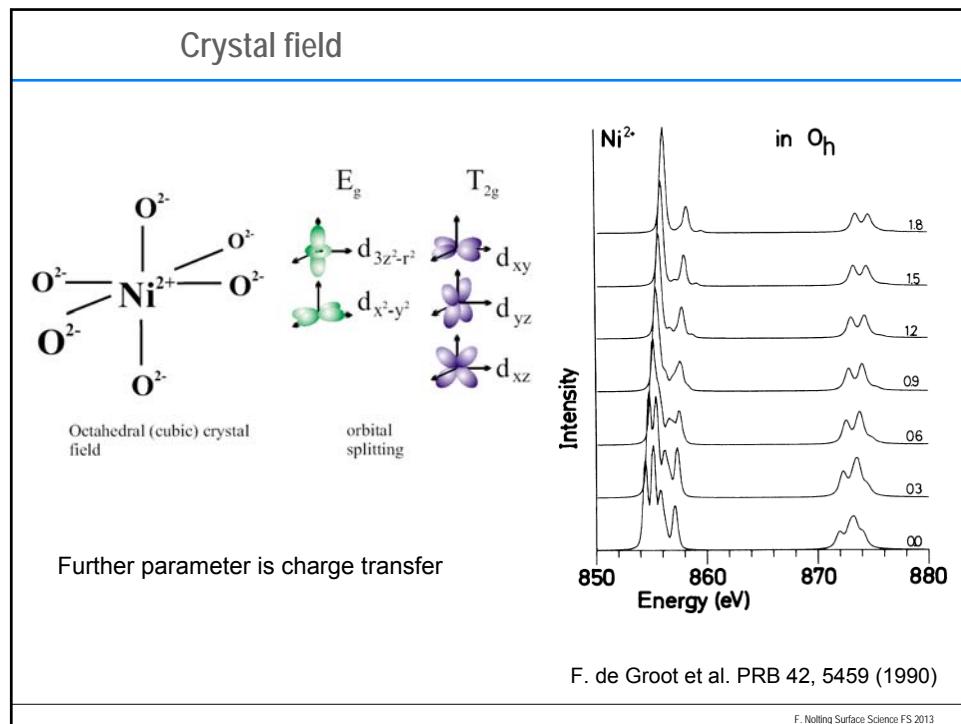
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Interactions – NiO (Ni 2+)



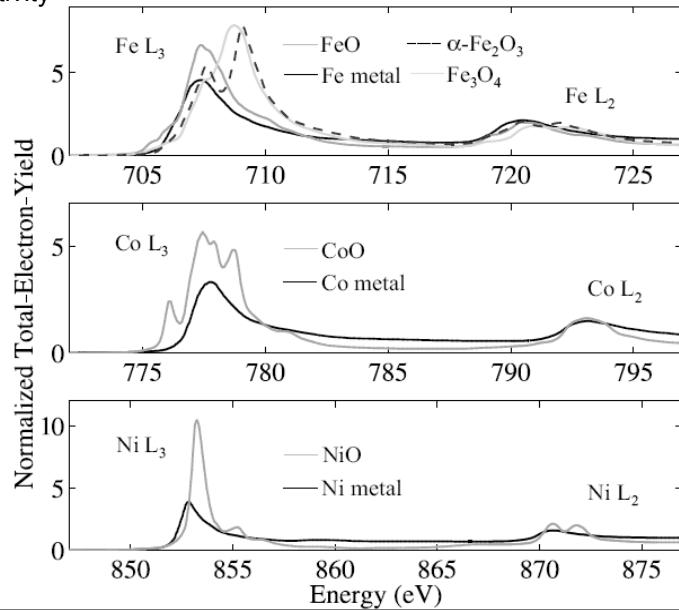
F. de Groot

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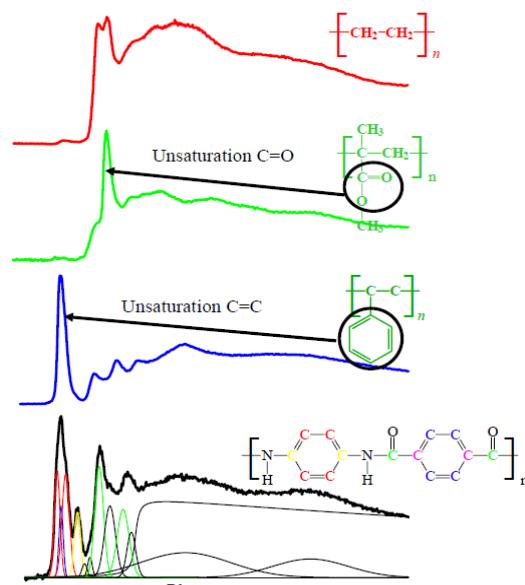


Complex but good for us

Chemical sensitivity

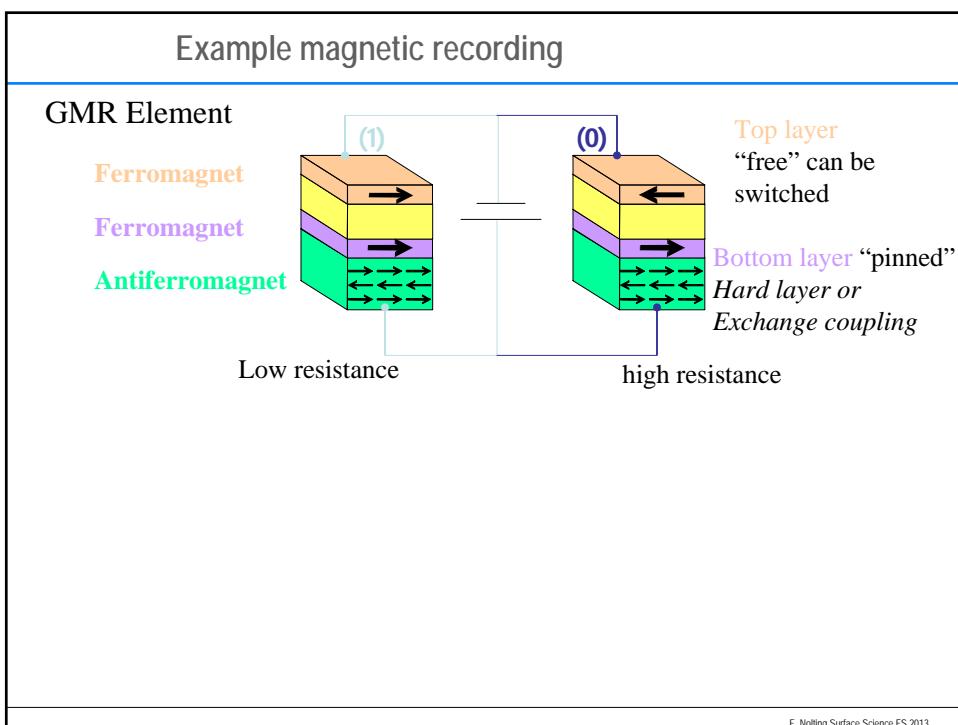
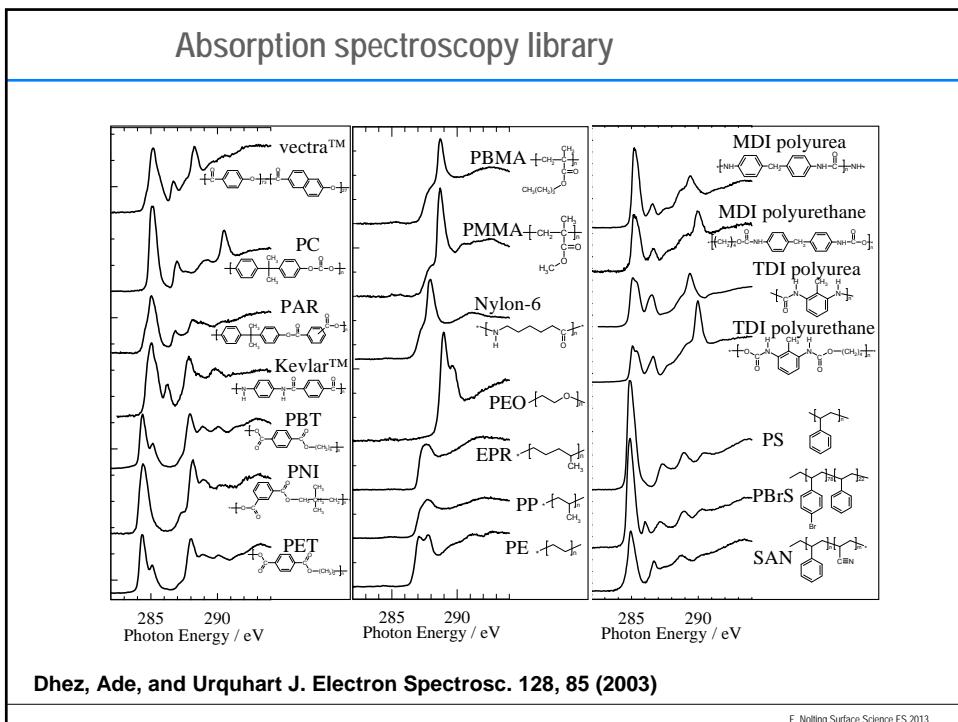


Footprint of complex bindings



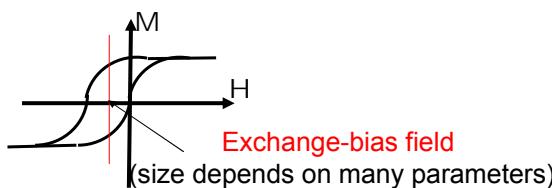
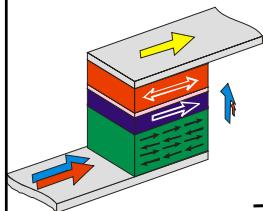
Courtesy Harald Ade

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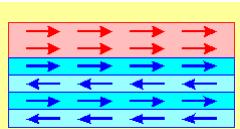
Exchange Bias

Unidirectional anisotropy in FM adjacent to AFM
discovered 1956 by Meiklejohn and Bean

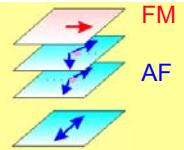


Effect of exchange bias is still poorly understood

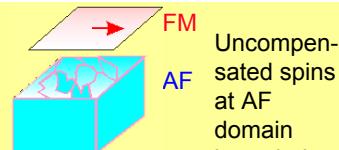
Ideal Interface Model



Spin Flop Model



Domain Size Model

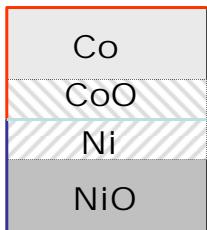
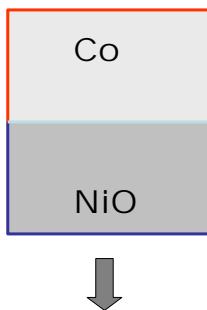


Uncompensated spins
at AF
domain
boundaries

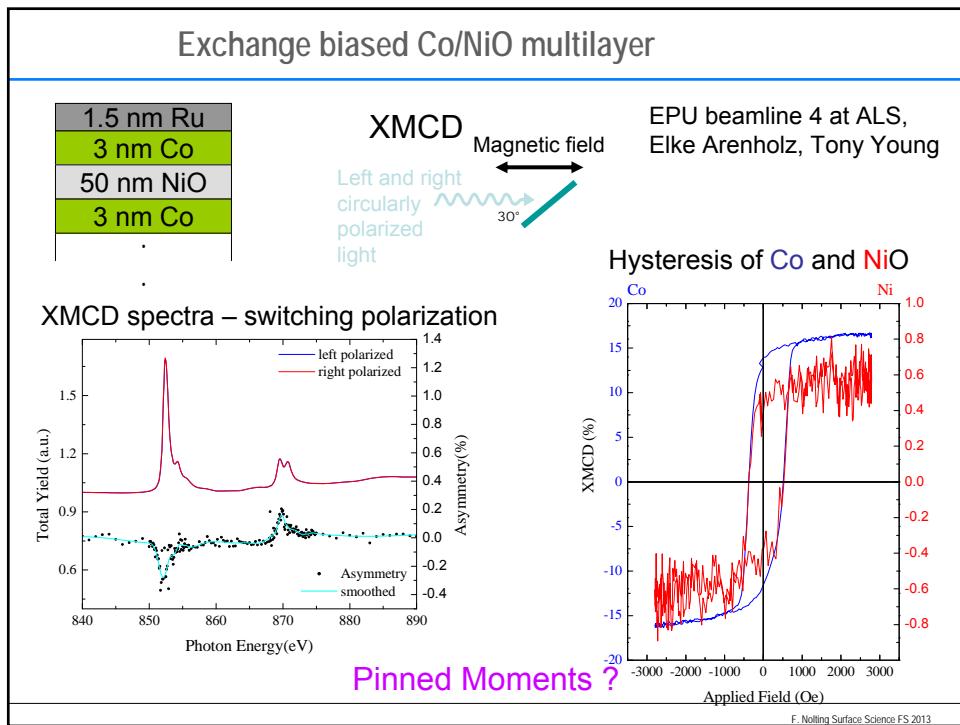
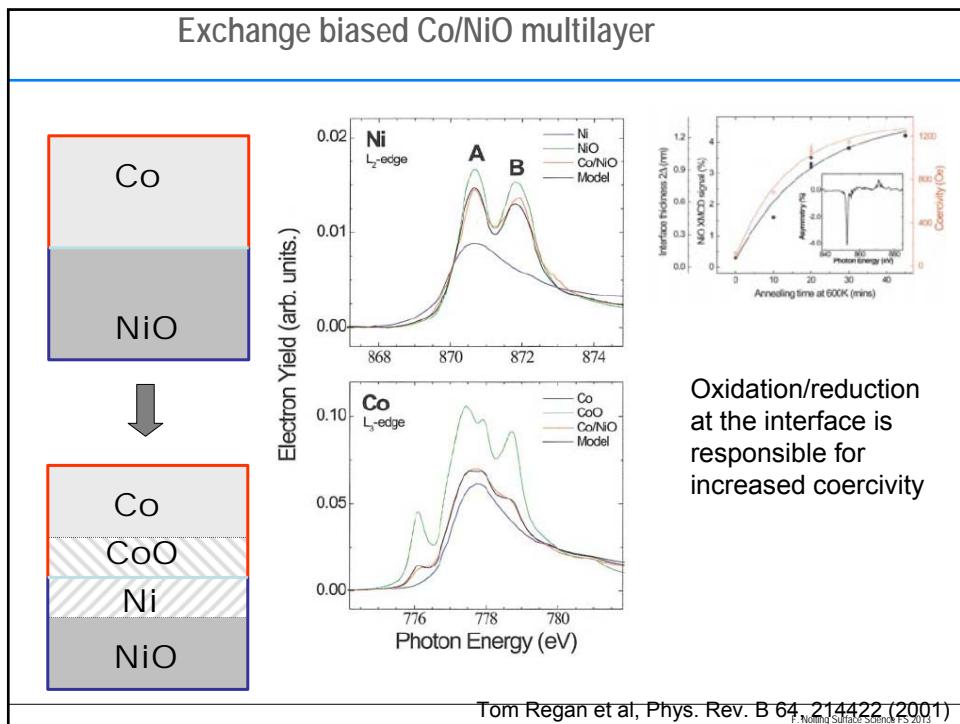
Reviews: A.E. Berkowitz and K. Takano, J. Magn. Magn. Mater. 200(1-3), 552 (1999).
J. Nogues and I.K. Schuller, J. Magn. Magn. Mater. 192, 203 (1999)

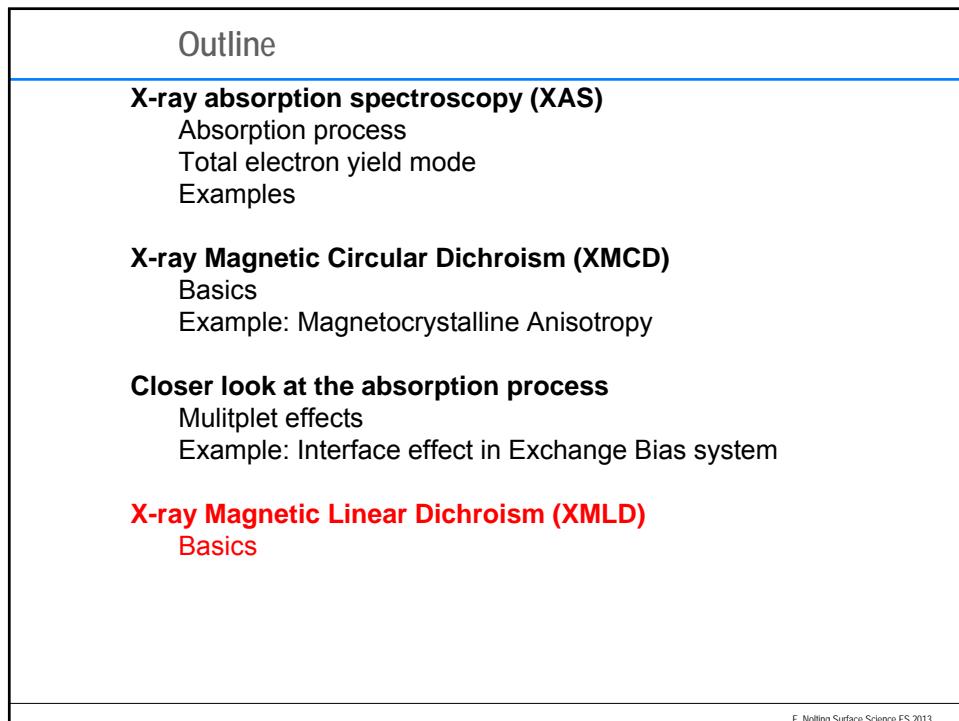
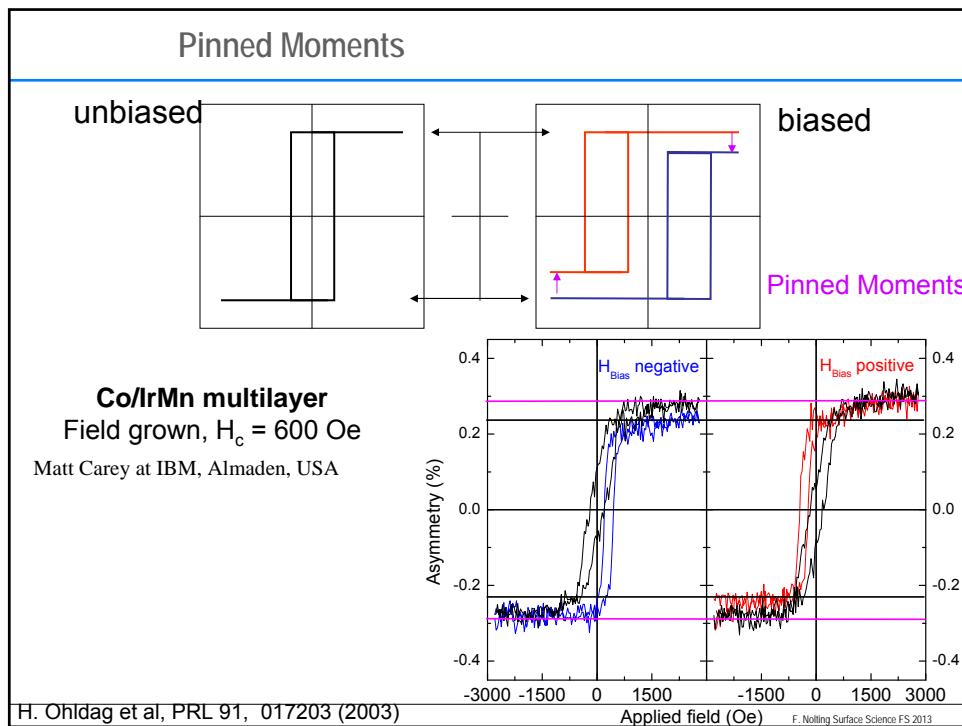
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Exchange biased Co/NiO multilayer



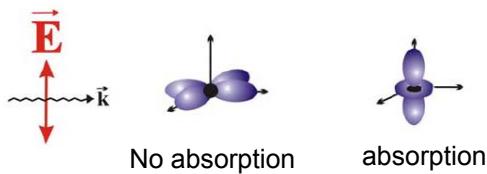
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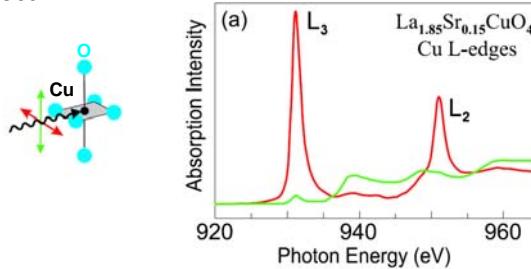


Interaction with linear light - charge

Excitation into 3d band



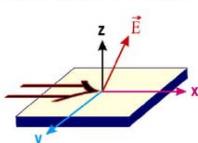
X-ray Natural linear dichroism
“search light effect”



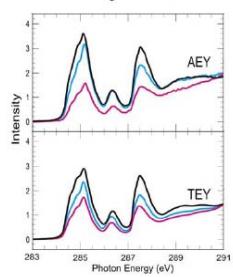
C.T. Chen et al PRL 68, 2543 (1998)

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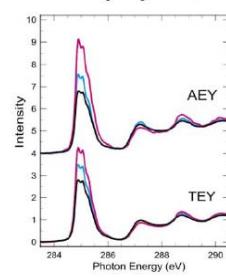
Polarization Dependent NEXAFS Probes Bond Anisotropy at Surface



Polyimide

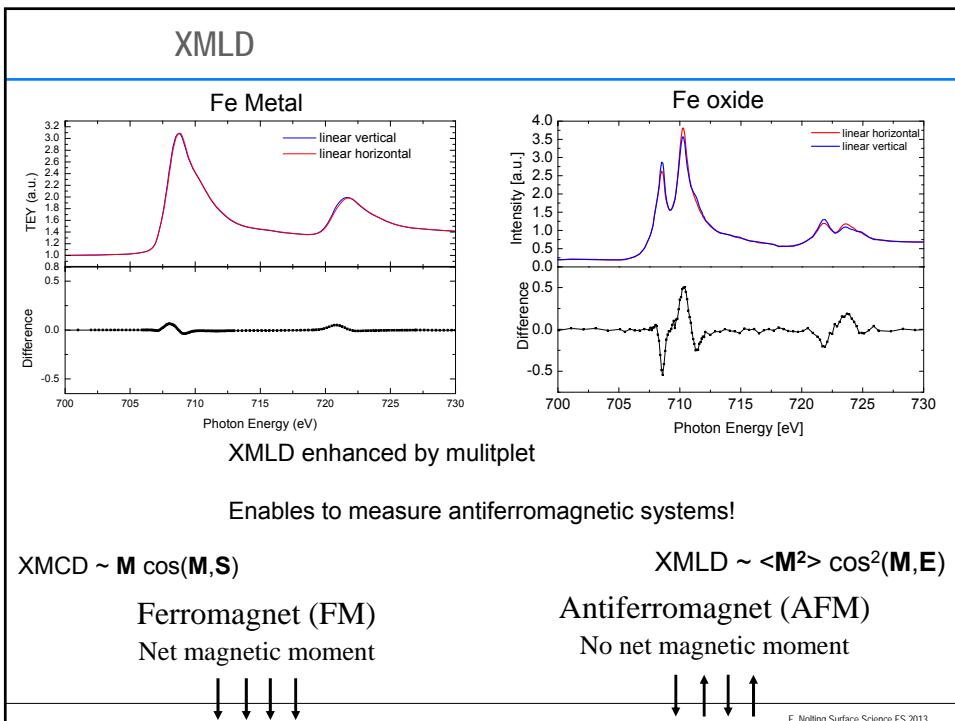
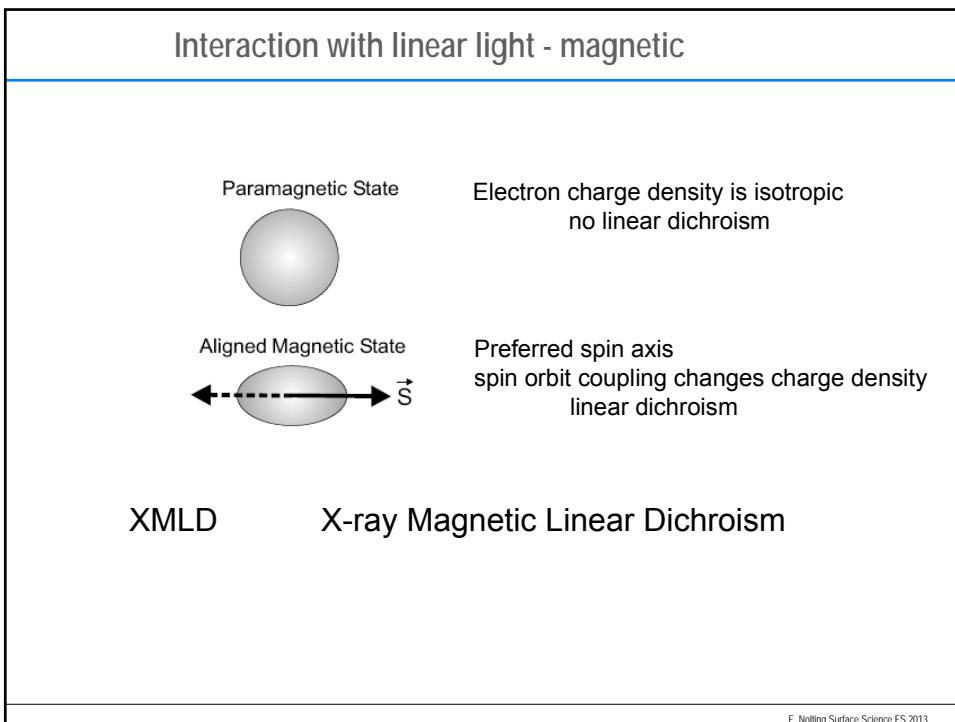


Polystyrene



J. Stöhr et al., Science 292, 2299 (2001)

Science FS 2013



Some pioneering papers on XMCD and XMLD

Strong magnetic dichroism predicted in the M_{4,5} X-ray absorption spectra of magnetic rare-earth materials,
B.T. Thole, G. van der Laan, and G.A. Sawatzky, Phys. Rev. Lett. 55, 2086 (1985).

Experimental proof of magnetic x-ray dichroism,
G. van der Laan, B.T. Thole, G.A. Sawatzky, J.B. Goekoop, J.C. Fuggle, J.-M. Esteva, R. Karnataka, J.P. Remeika, and H.A. Dabkowska, Phys. Rev. B 34, 6529 (1986).

Absorption of Circularly Polarized X-rays in Ion,
G. Schütz, W. Wagner, W. Wilhelm, P. Kienle, R. Zeller, R. Frahm, G. Materlik, Phys. Rev. Lett. 58, 737 (1987).

Soft X-ray magnetic circular dichroism at the L_{2,3} edges of nickel,
C.T. Chen, F. Sette, Y. Ma, and S. Modesti, Phys. Rev. B 42, 7262 (1990).

X-ray circular dichroism as a probe of orbital magnetization,
B.T. Thole, P. Carra, F. Sette, and G. van der Laan, Phys. Rev. Lett., 1943 (1992).

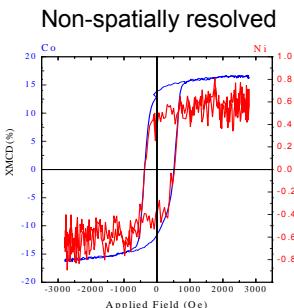
Magnetic X-ray dichroism – general features of dipolar and quadrupolar spectra,
P. Carra, H. König, B.T. Thole, and M. Altarelli, Physica B 192, 182 (1993).

Determination of Spin- and Orbital-Moment Anisotropies in Transition Metals by Angle-Dependent X-Ray Magnetic Circular Dichroism,
J. Stöhr, H. König, Phys. Rev. Lett. 75, 3748 (1995)

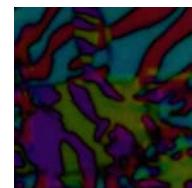
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Conclusions

Polarization depend soft X-ray absorption spectroscopy is a powerful tool to study elemental resolved ferromagnetic and antiferromagnetic thin films and interfaces



spatially resolved ...next time



The technique

is sensitive to

- elemental composition
- chemical bonds
- structural parameters
- electronic structure
- magnetic properties

has a very broad range of application

- Magnetic multilayers
- Diluted systems
- Nanoparticles
- Molecular magnets

...

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