

Surface Science lecture

Di, 09.03.2010 Introduction, Concepts Samples and Structure (Thomas)
Di, 16.03.2010 Electron Diffraction Methods, in particular RHEED, LEED (Bert Mueller)
Di, 23.03.2010 Adsorption / Desorption (Thomas Jung)

Di, 30.03.2010 X-ray Absorption Spectroscopy (F. Nolting)
Di, 06.04.2010 Surface Magnetism XMCD / PEEM (F. Nolting)

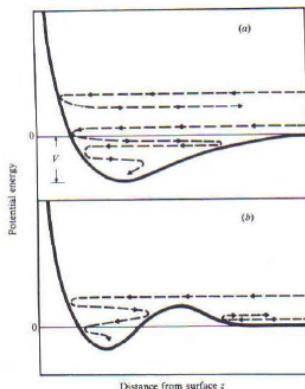
Di, 13.04.2010 Diffusion and Growth (Enrico Gnecco)
Di, 20.04.2010 Electronic Properties and Surface Electron Spectroscopies
XPS/UPS, Auger, ARPES; (Nirmalya Ballav???)
Di, 27.04.2010 Local Probes and Experiments I, STM, Inelastic tunneling and STS (Silvia Schintke)
Di, 04.05.2010 Local Probes and Experiments II, AFM FIM (E. Meyer???)
Di, 11.05.2010 Surface Optics, Kelvin Probe (Thilo Glatzel)
Di, 18.05.2010 Applications of Surface Science in Industry (M. de Wild)
Di, 25.05.2010 Schlusspruefung (Thilo, Christian, Cristian, Enrico, Thomas)
Di, 01.06.2010 Exkursion to e.g. IBM Zurich Research Laboratory (Nanolab, NJ-Lab and Thomas)

Di, 30.03.2010 X-ray Absorption Spectroscopy (F. Nolting)
Di, 06.04.2010 X-ray Microscopy (F. Nolting)

Both with an emphasis of magnetism

Repetition

Adsorption/Desorption



Was geschieht mit einem Atom/Molekül welches aus der Gasphase auf der Oberfläche landet?

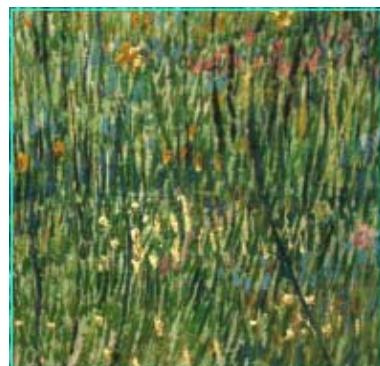
(a) Je nach Energie:
Repulsion und Reflexion ins Vakuum oder Umwandlung der kinetischen Energie in Vibrations- und Rotationsfreiheitsgrade und schliesslich Adsorption (Physisorption oder Chemisorption)

(b) Auf gewissen Flächen können auch niedrigerenergetische Teilchen reflektiert werden. Bzw. Verweilen in einem Zwischenzustand.

Repetition

adsorption / desorption ‘take home’ message

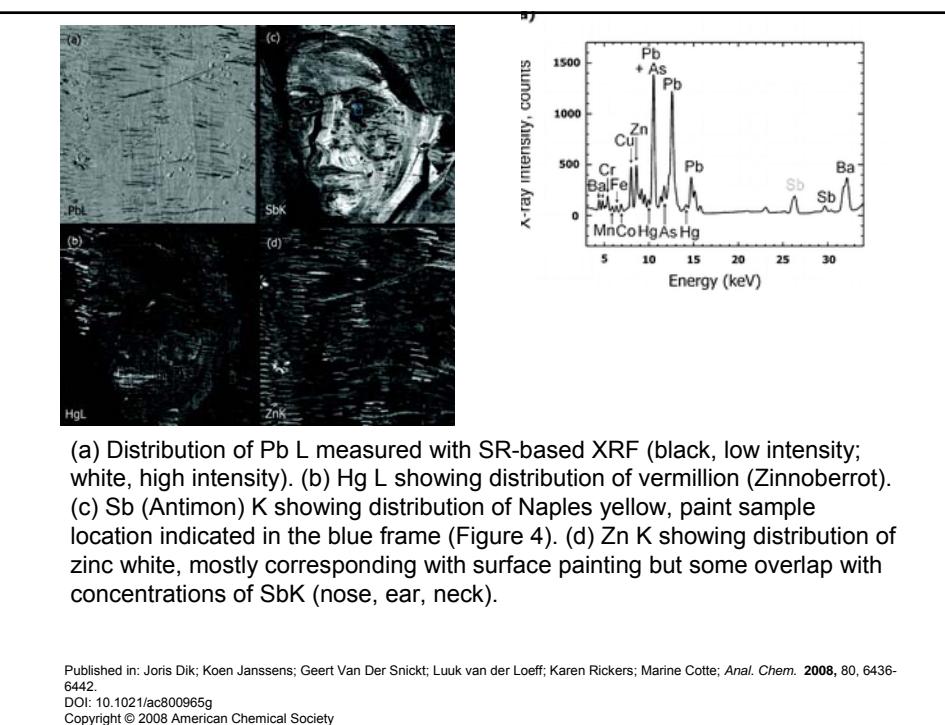
- Complex multi – stage process
- T variation affects relaxation
- Important for the formation of interfaces and thin films
- Important for catalysis
- Important for the analysis of interface bonding
- Adsorption can be treated like a chemical reaction:



The 1887 floral painting by van Gogh, “Patch of Grass”.



The 1887 floral painting by van Gogh, "Patch of Grass". XRF analysis within the area bounded by the blue box revealed details of a hidden face of a peasant woman from an earlier painting (ca. 1884 – 1885). (b) A tritonal reconstruction from Sb (yellow-white) and Hg (vermillion) elemental distribution maps of the hidden portrait. From Dik et al., *Anal. Chem.* 80 6436 (2008).



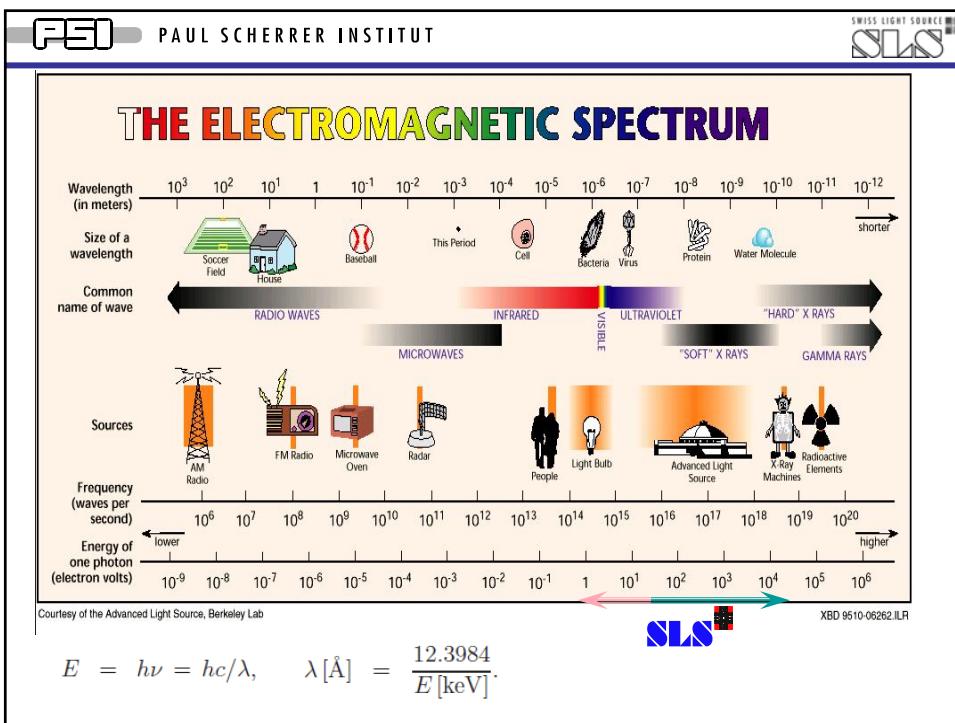
Some things are hidden behind the surface ...

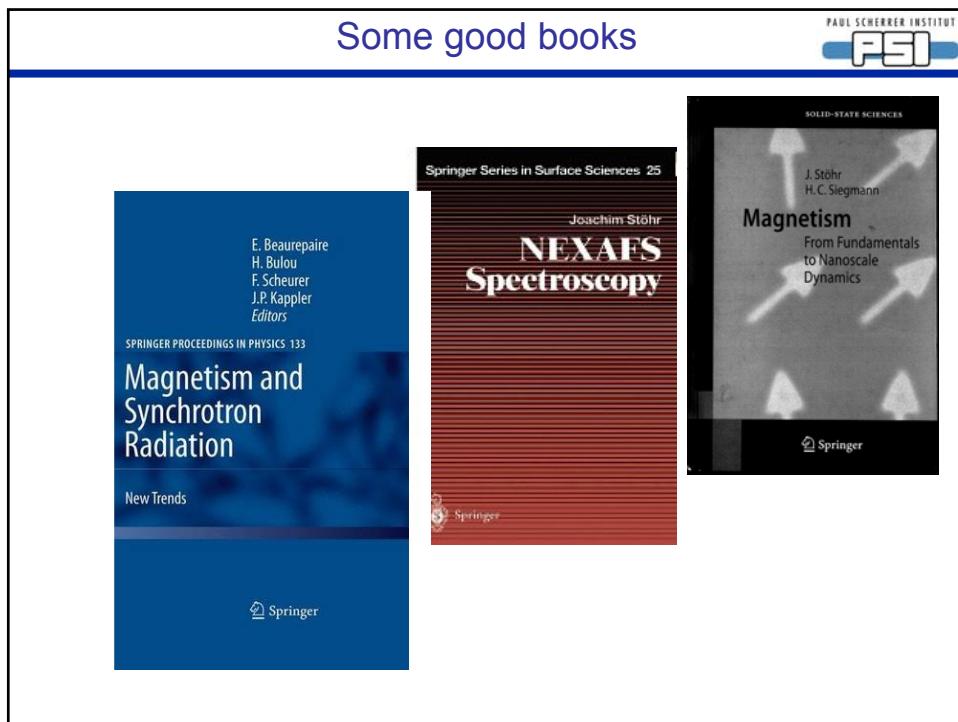
what is surface science?

interaction at the surface/interface getting more and more important
combine information

Visible light gives restricted information

X-ray are an excellent extension

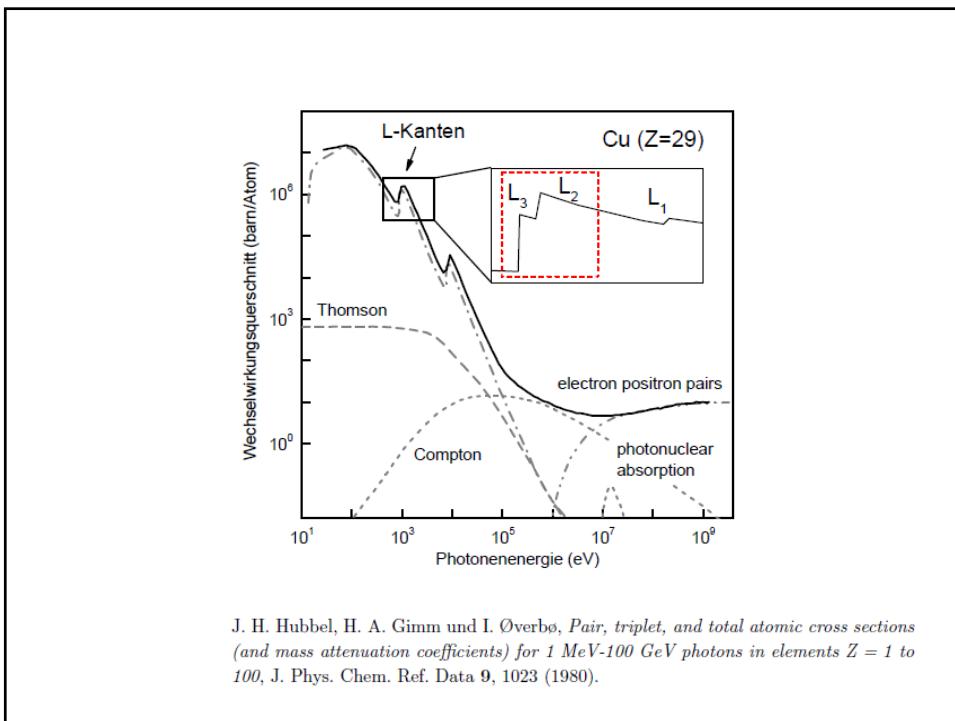
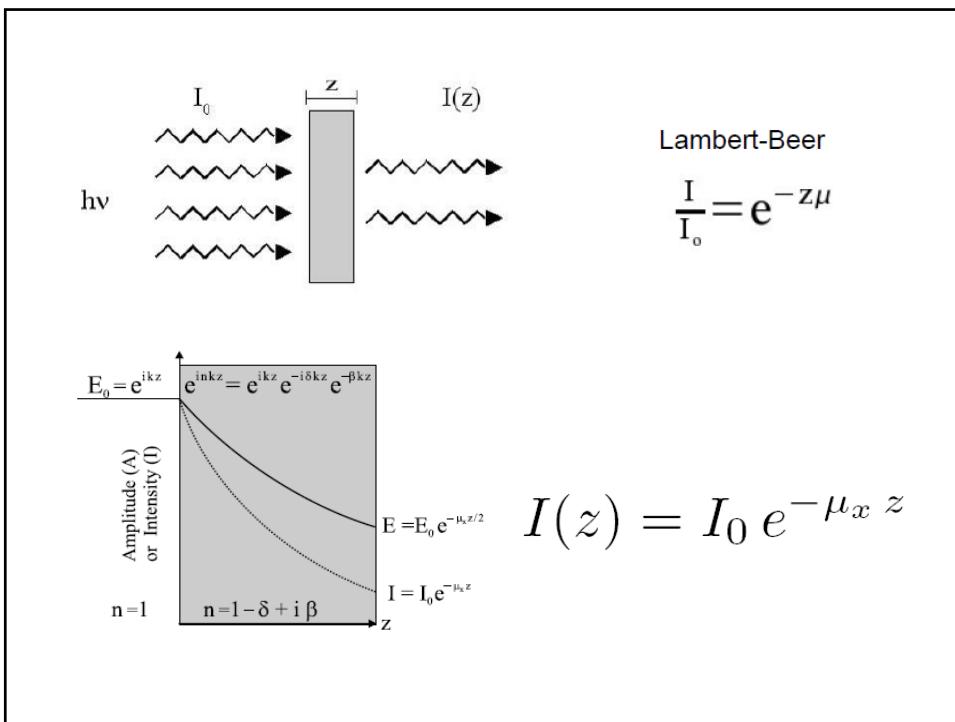


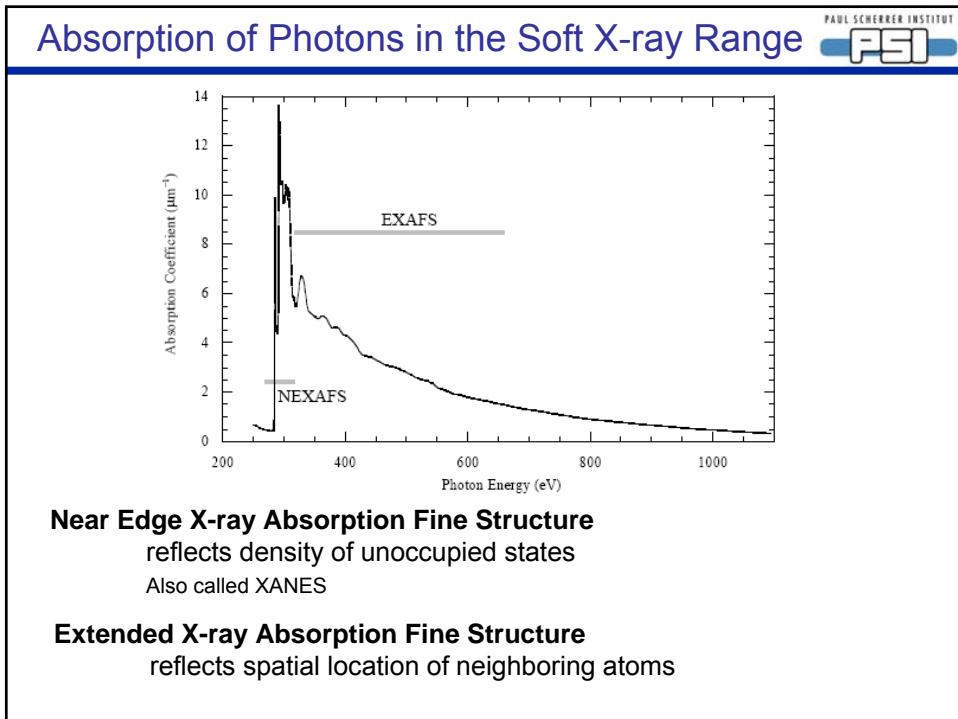
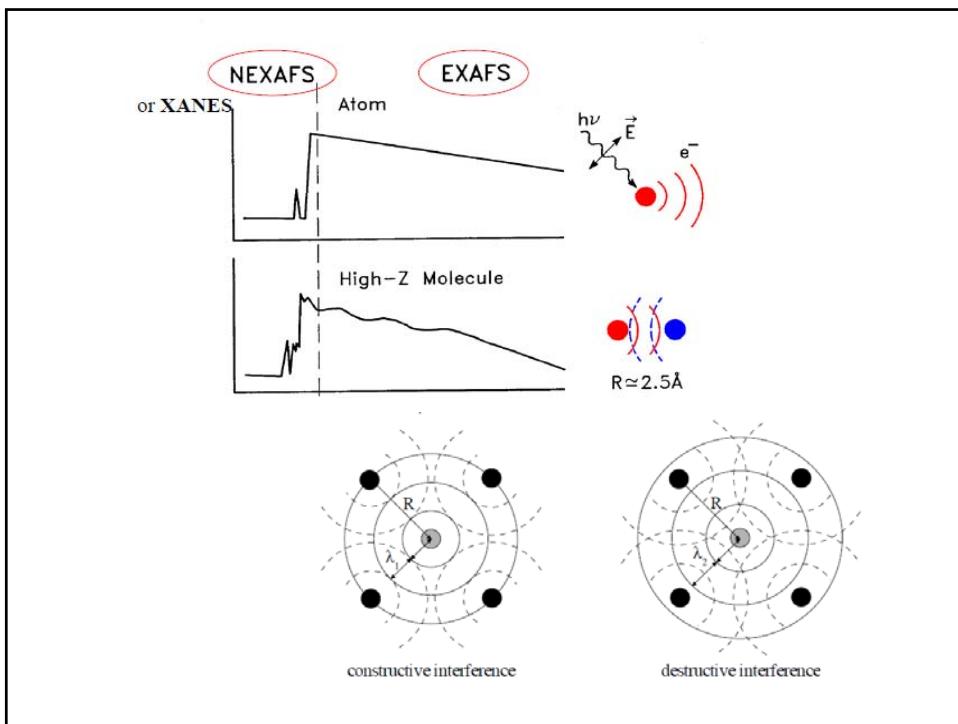


Outline

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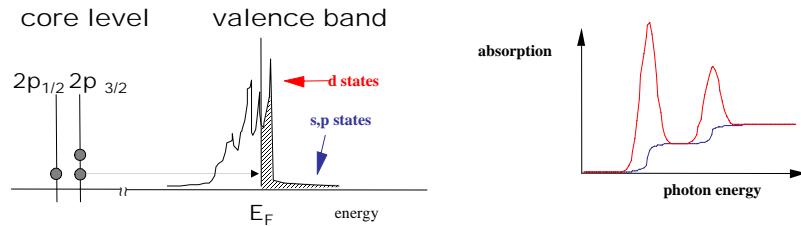
- X-ray absorption spectroscopy (XAS)**
 - Absorption process
 - Total electron yield mode
 - Examples
- X-ray Magnetic Circular Dichroism (XMCD)**
 - Basics
 - Example: Magnetocrystalline Anisotropy
- Closer look at the absorption process**
 - Multiplet effects
 - Example: Interface effect in Exchange Bias system
- X-ray Magnetic Linear Dichroism (XMLD)**
 - Basics





Interaction of electromagnetic wave with charge

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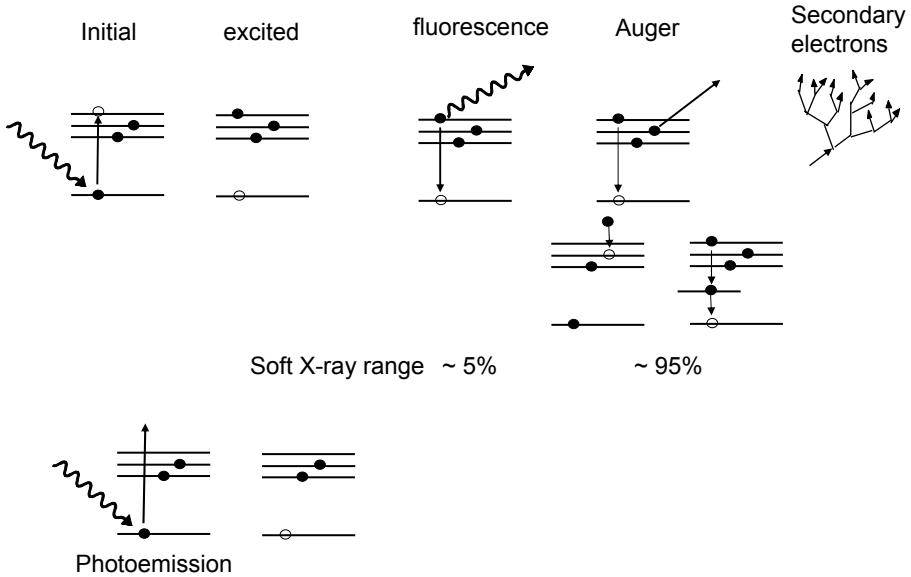
$$\text{Absorption} \sim \frac{\text{Transition matrix}}{\text{Final state Initial state}} \cdot \text{Density of final states}$$

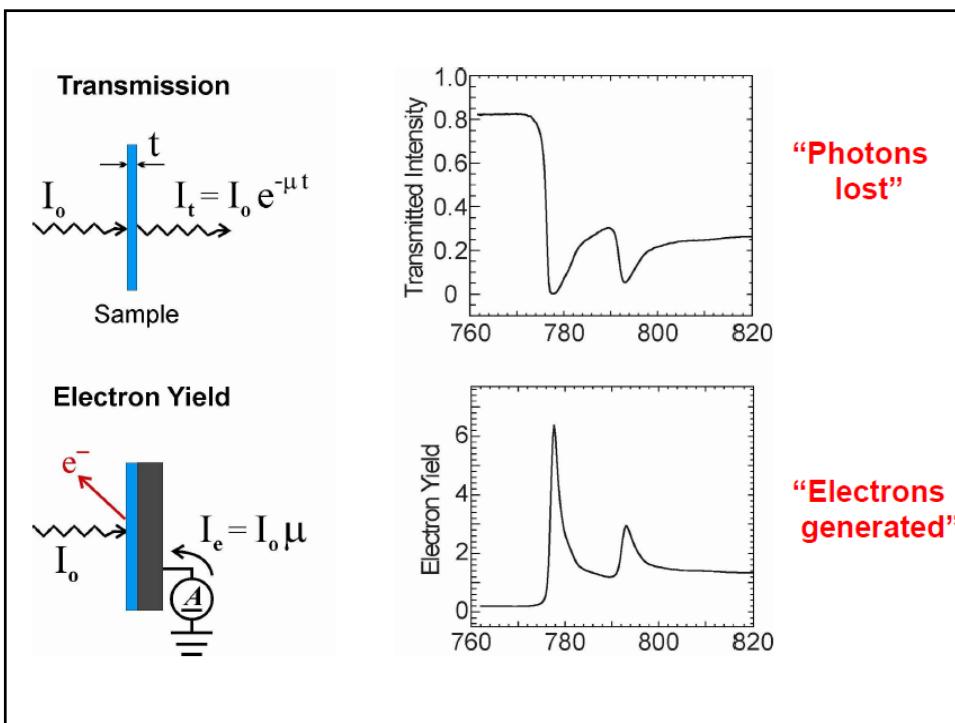
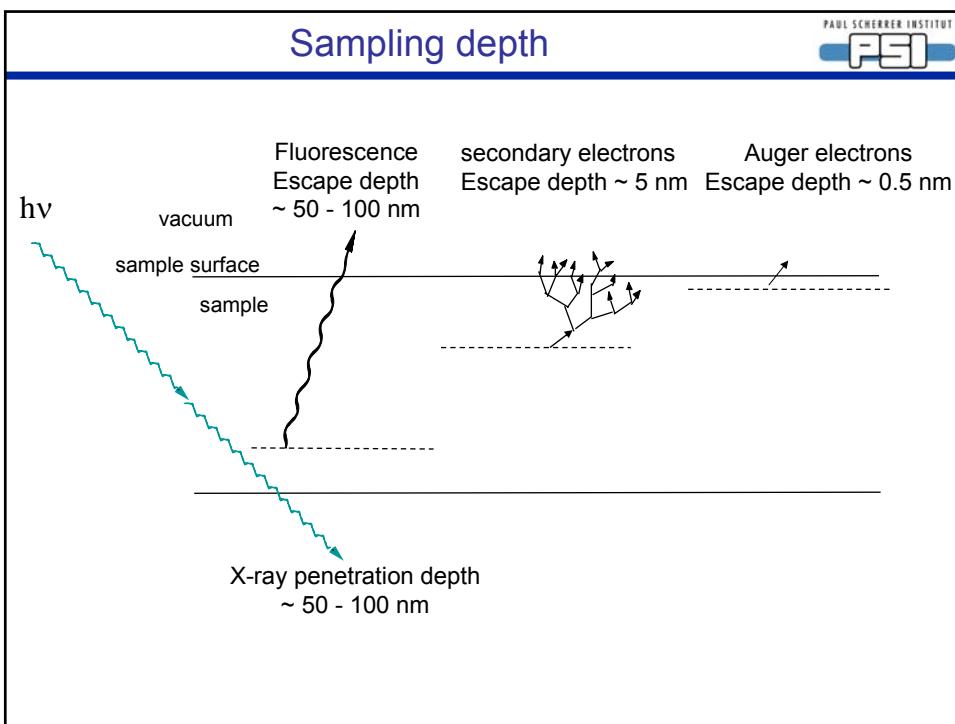
Fermi's golden rule
in dipole approximation
wavelength is large compared to charge
1000 eV corresponds to 1.2 nm
2p core radius is about 0.01 nm

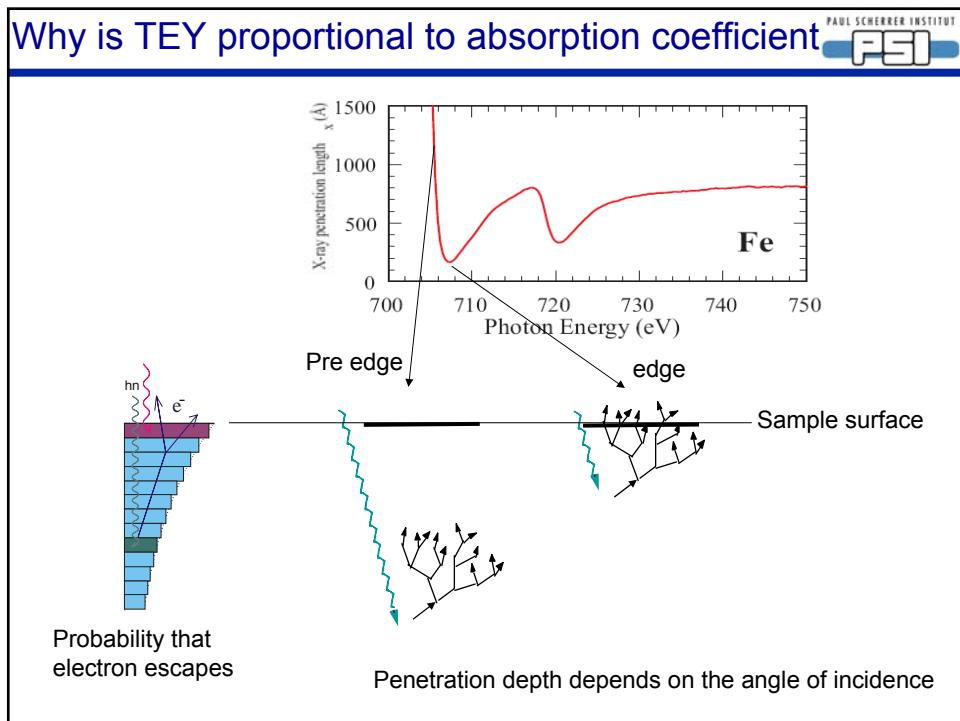
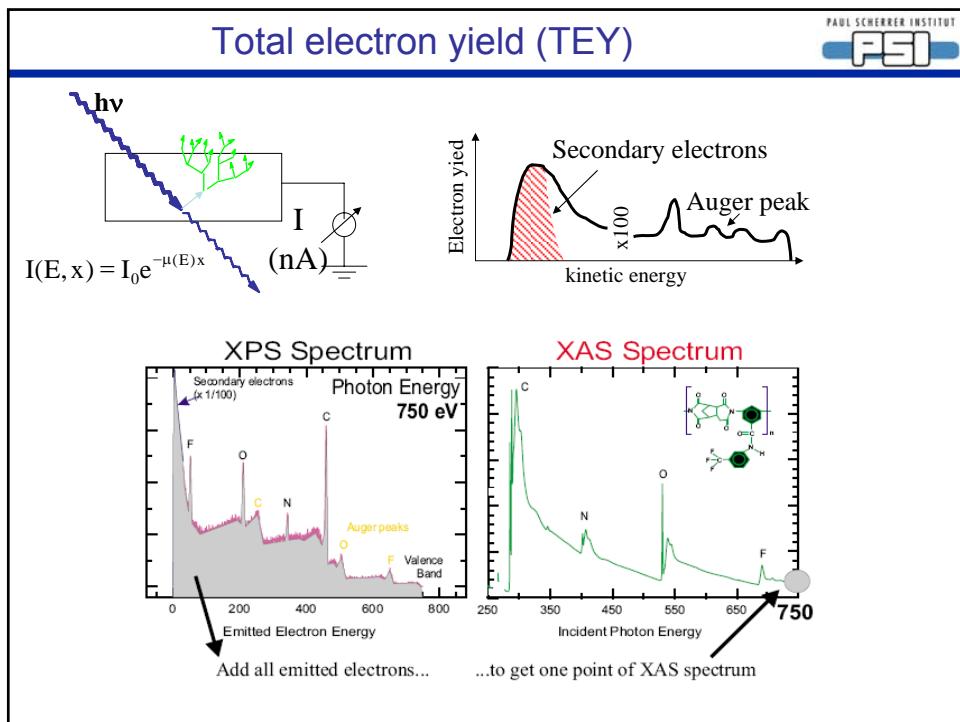
$$\mu \propto |\langle f | e \cdot p | i \rangle|^2 \rho(E)$$

Decay channels

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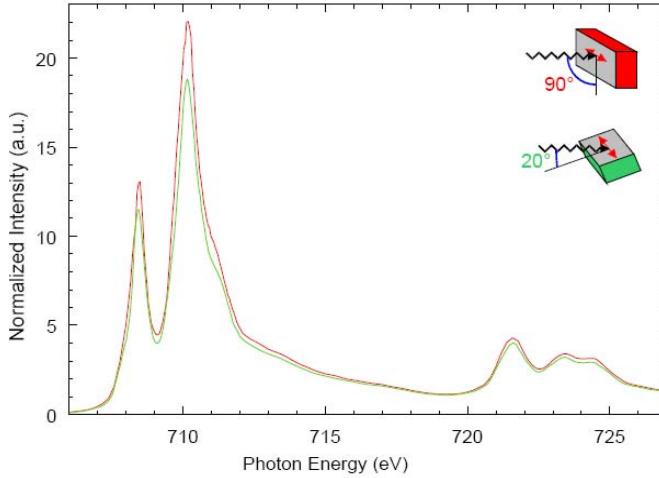






Saturation Effects in TEY Detection

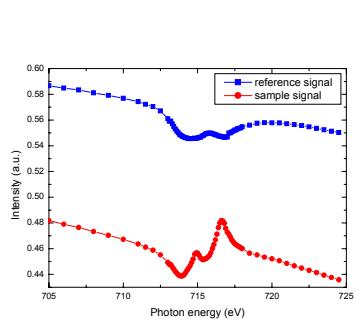
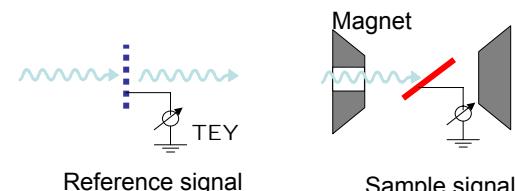
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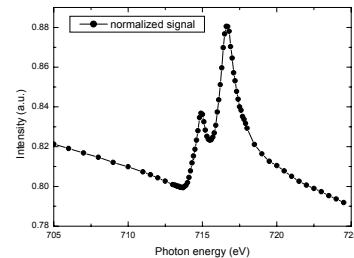
J. Lüning et al, PRB 67, 214433 (2003)

How do we measure

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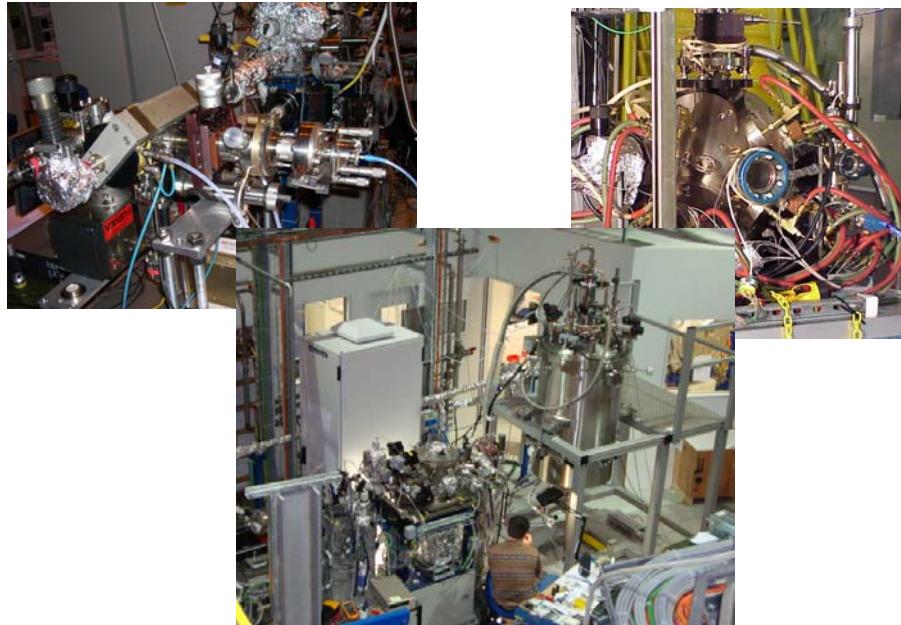


$$\frac{\text{Sample}}{\text{Reference}} = \text{Norm}$$



How do we measure

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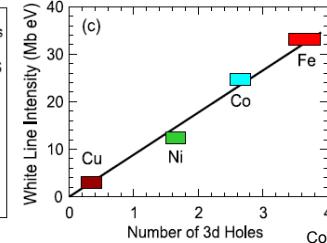
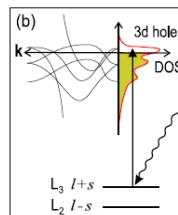
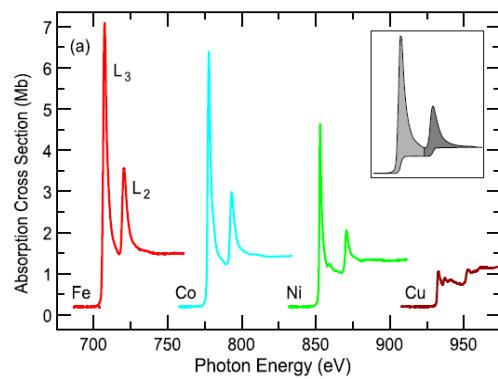


Determine number of 3d holes

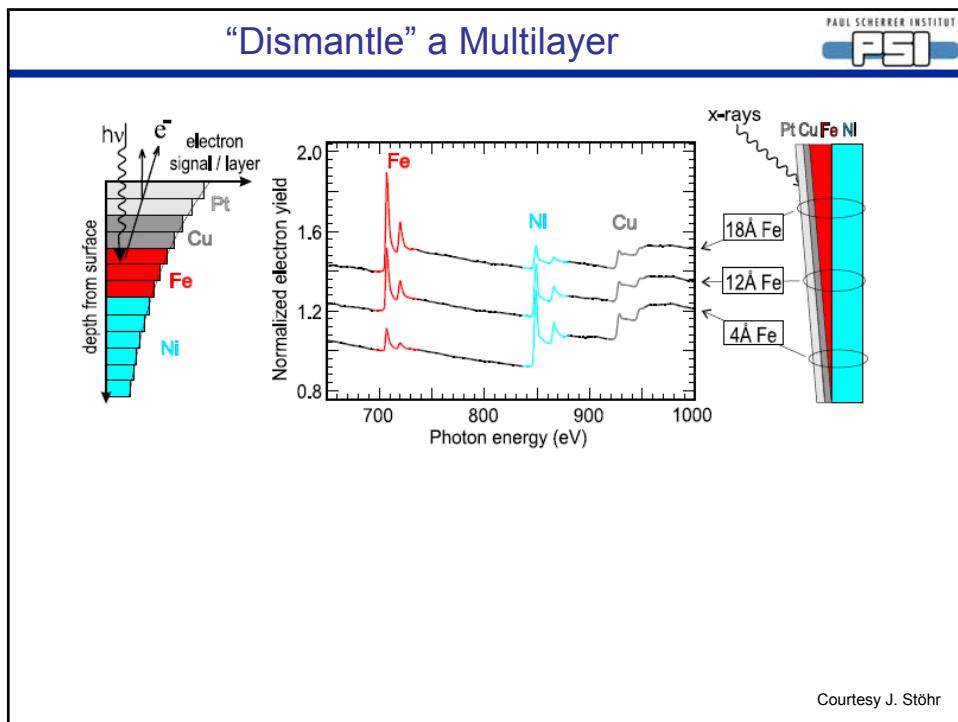
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Charge sum rule

Integrated intensity
is proportional to
number of empty
valence states



Courtesy J. Stöhr



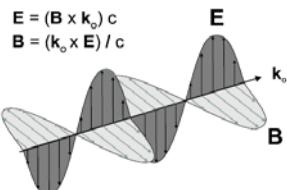
Outline

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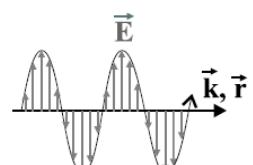
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Polarized Photons

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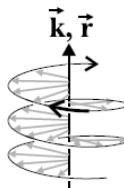


Linear polarization

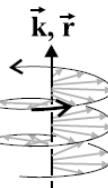


Left circular polarization

space



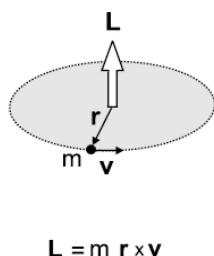
Right circular polarization



Polarized Photons

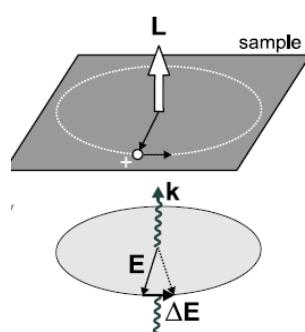
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Angular momentum of orbiting mass

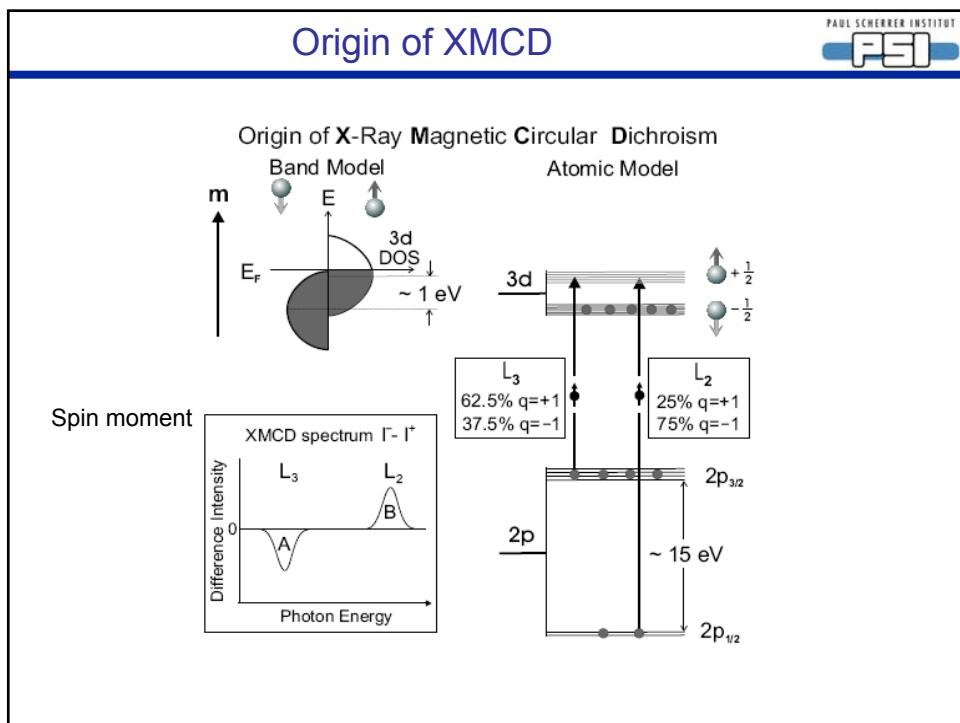
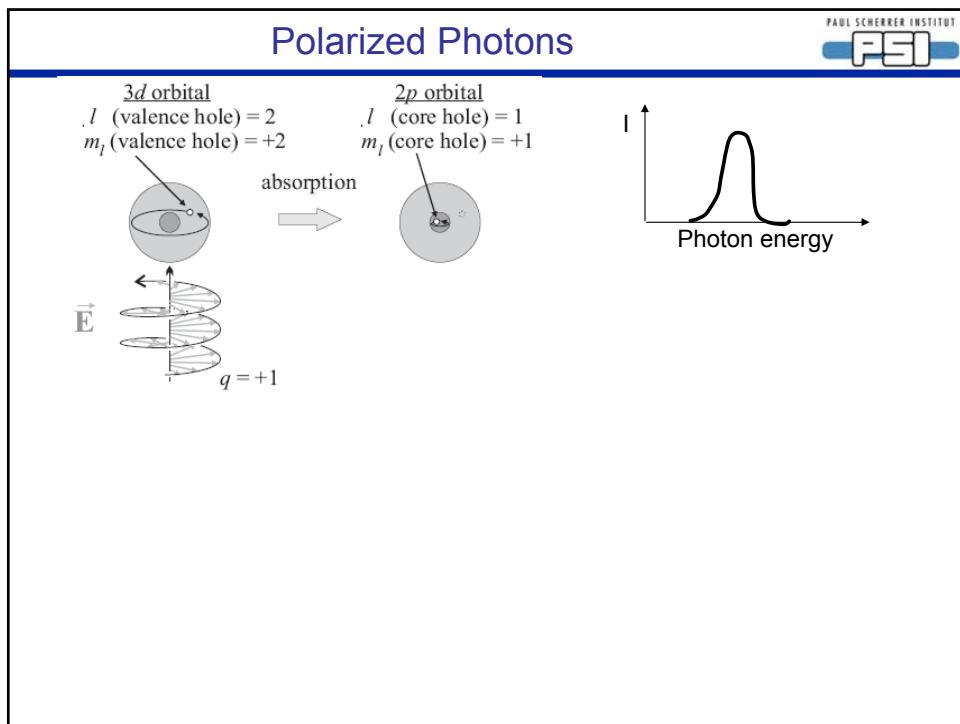


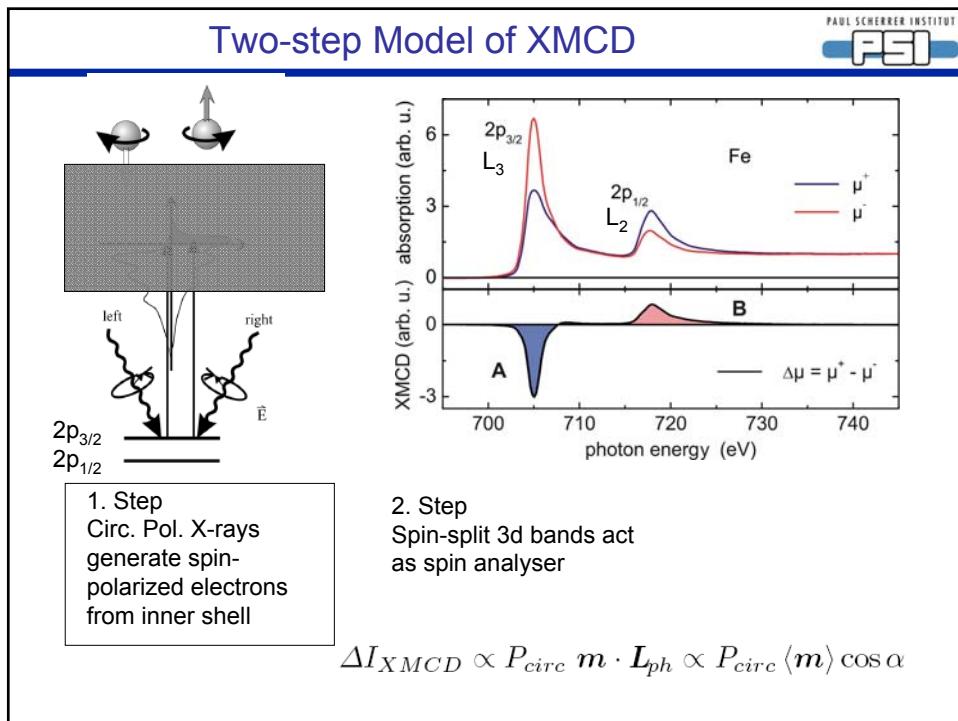
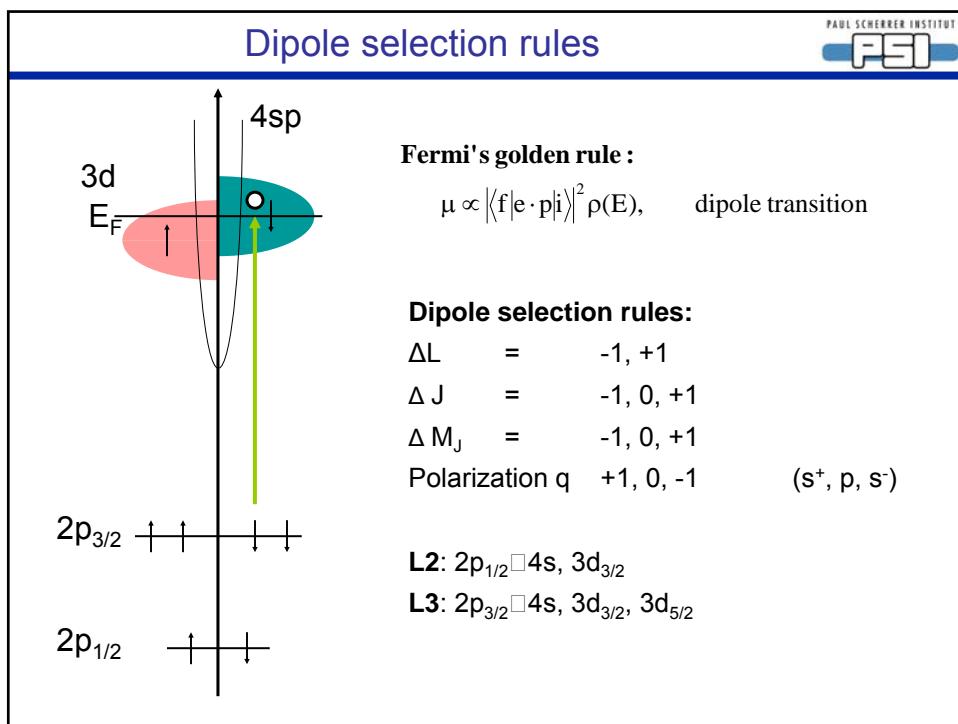
$$L = m r \times v$$

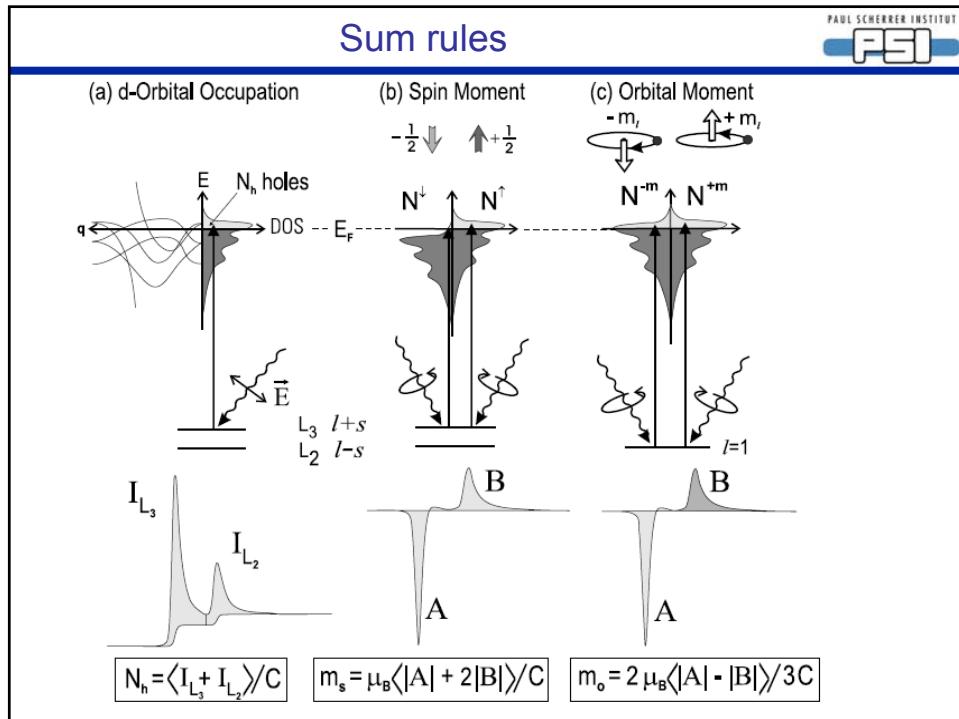
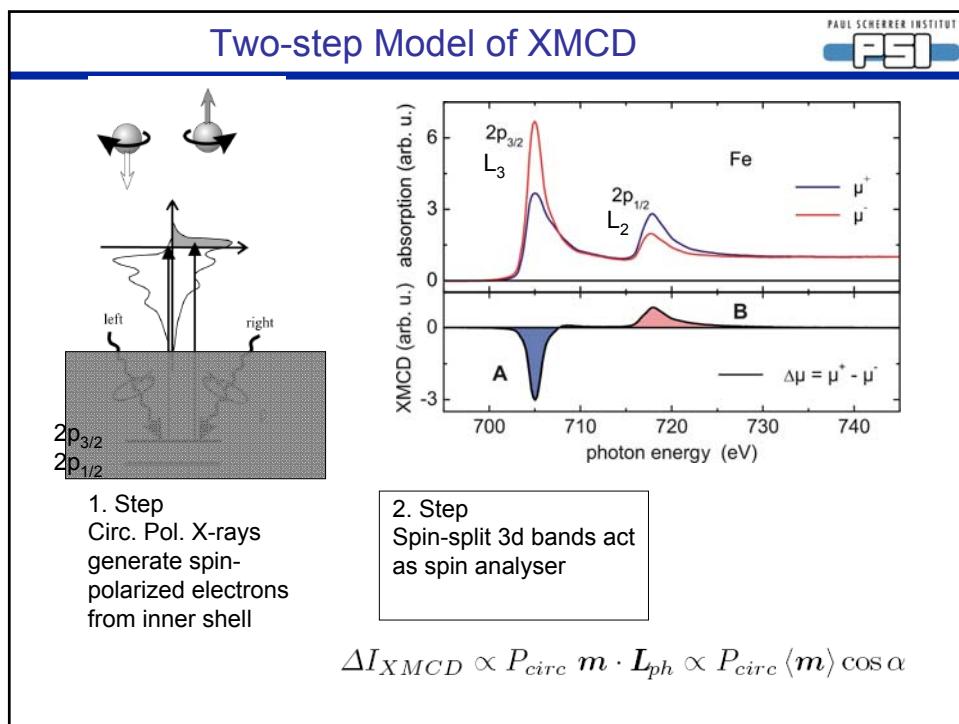
Photon angular momentum



Angular momentum conservation







... more complex ... of course

$$m_{\text{orb}} = - \frac{4 \int_{L_3+L_2} (\mu_+ - \mu_-) d\omega}{3 \int_{L_3+L_2} (\mu_+ + \mu_-) d\omega} (10 - n_{3d})^{\leftarrow}, \quad (1)$$

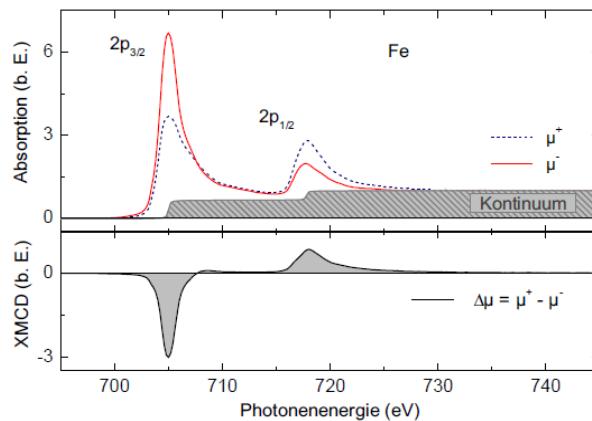
$$m_{\text{spin}} = - \frac{6 \int_{L_3} (\mu_+ - \mu_-) d\omega - 4 \int_{L_3+L_2} (\mu_+ - \mu_-) d\omega}{\int_{L_3+L_2} (\mu_+ + \mu_-) d\omega} \\ \times (10 - n_{3d}) \left(1 + \frac{7 \langle T_z \rangle}{2 \langle S_z \rangle} \right)^{-1}, \quad (2)$$

$\langle T_z \rangle$ is the expectation value of the intra-atomic magnetic dipole operator, accounting for a possible asphericity of the spin density distribution.

effective spin magnetic moment

$$\mu_S^{\text{eff}} = \mu_S + 7\mu_T$$

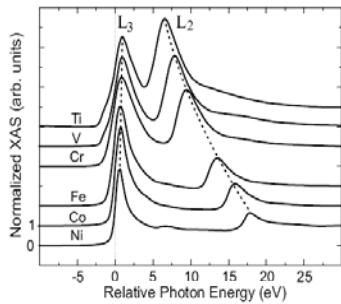
Only contribution to the resonance absorption should be considered



Taken from PhD Thesis Armin Kleibert, 2005

... more complex ... of course

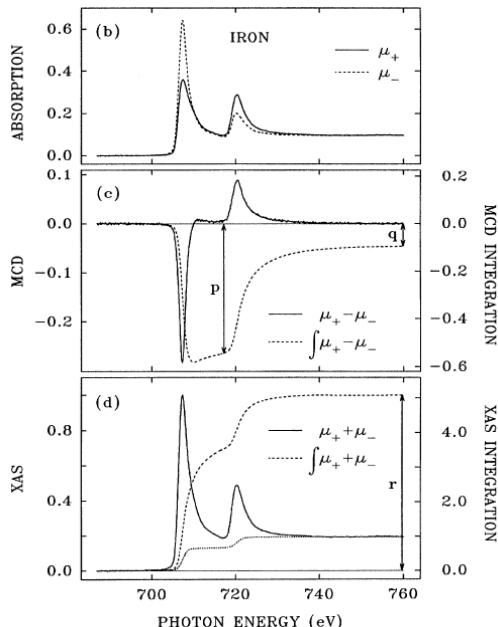
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L_3 and L_2 must be separated

calculating

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Normalize Spectra
Corrected for
incident angle
Polarization degree
saturation effects

MCD
difference
integrate

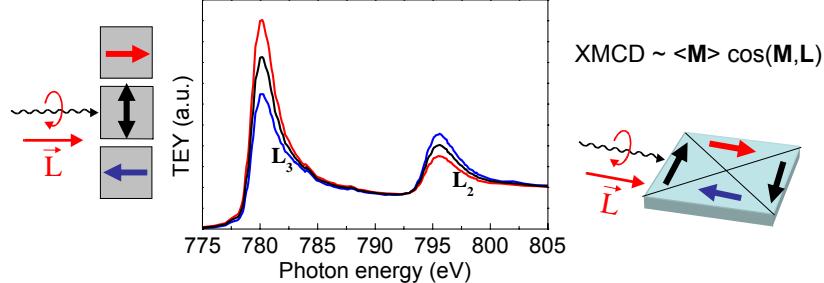
XAS
sum spectra
correct background
non resonant

Calculate/neglect
 $\langle T_z \rangle$
 $\langle S_z \rangle$

C.T. Chen et al. PRL 75(1), 152 (1995)

X-ray Magnetic Circular Dichroism (XMCD)

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Photons

- no mass
- no spin
- angular momentum

Magnetism 3d metals

- small orbital moment
- large spin moment

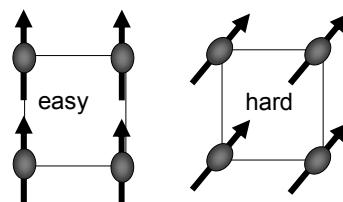
XMCD in action

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Study Magneto-crystalline anisotropy

Magnetic Anisotropy
preferential magnetization along axes
easy / hard axis

(magneto-crystalline anisotropy)

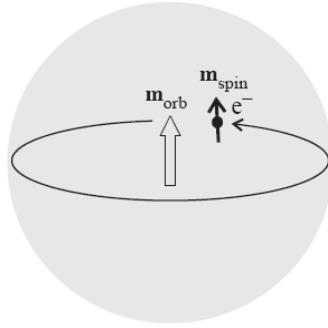


The magneto-crystalline anisotropy is the energy that it takes to rotate the magnetization from the “easy” direction into the “hard” direction

J. Stöhr, Jmmm 200 (1999) 470 – 497
Reiko Nakajima PhD Thesis 1998

Magneto-crystalline anisotropy

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Spin moment $\sim 1.5 \mu_B / \text{atom}$ isotropic

Orbital moment $\sim 0.1 \mu_B / \text{atom}$ isotropic/anisotropic

They interact via the spin-orbit coupling $L \cdot S$

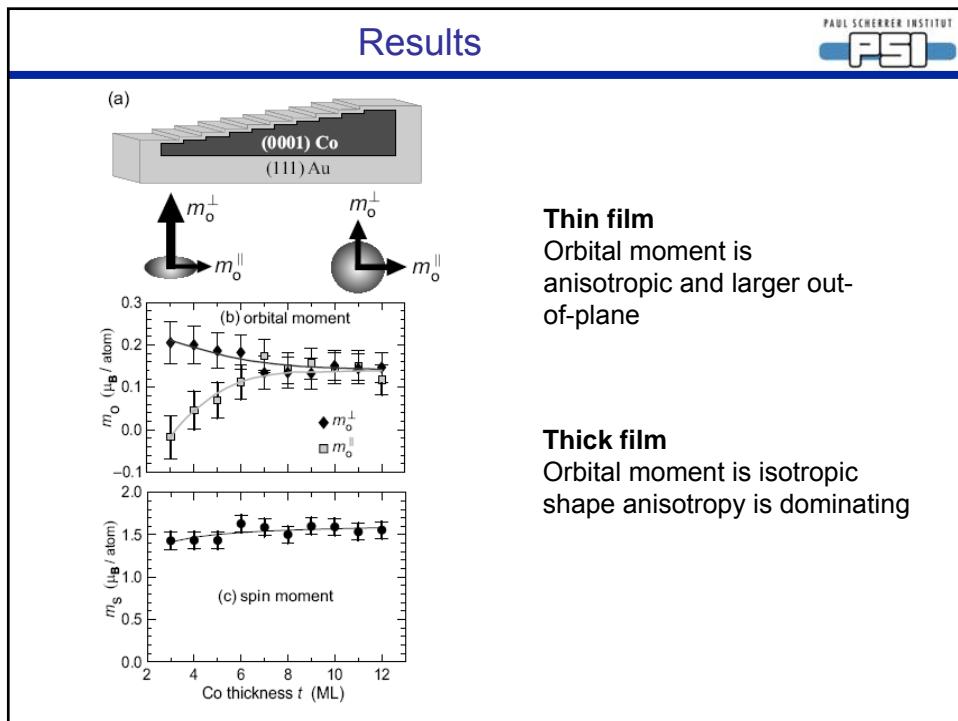
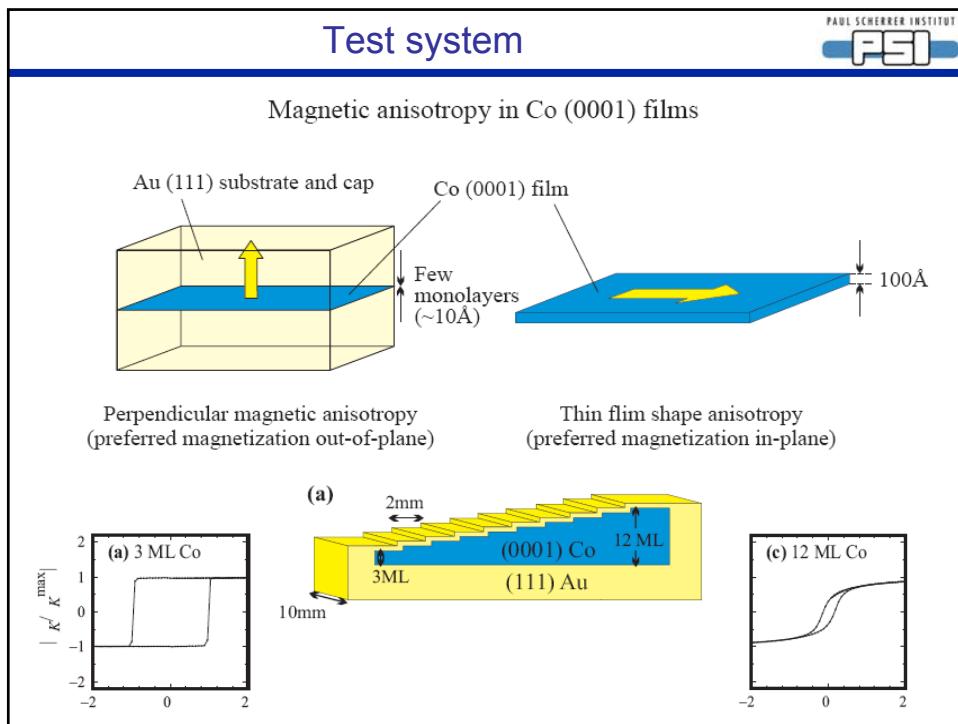
Bruno model

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The Bruno model states that the orbital moment is larger along the easy magnetization direction, and that the difference between the orbital moments along the easy and hard directions is proportional to the magneto-crystalline anisotropy

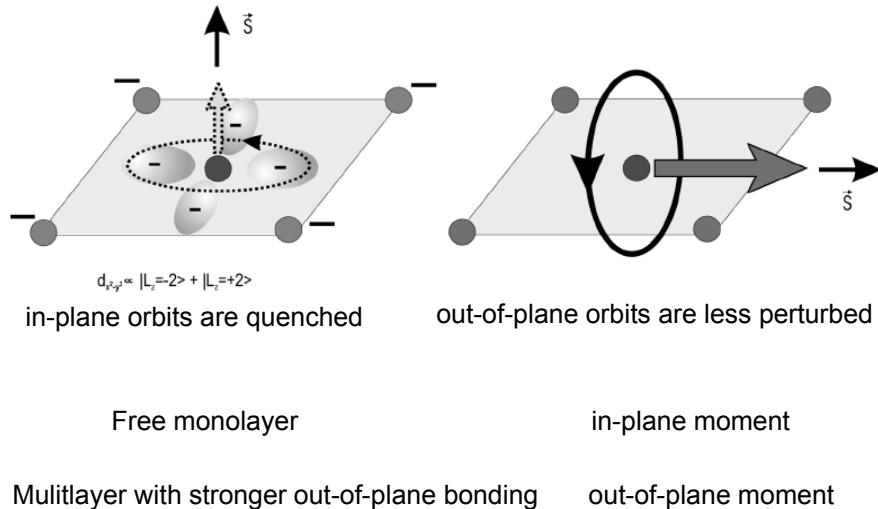
$$\Delta E_{so} = \zeta [\langle L \cdot S \rangle_{\text{hard}} - \langle L \cdot S \rangle_{\text{easy}}] = \frac{\zeta}{4\mu_B} (m_o^{\text{easy}} - m_o^{\text{hard}}) > 0$$

P. Bruno, PRB 39, 865 (1989)



Simple picture – Ligand fields

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Ferromagnetism in one-dimensional monatomic metal chains

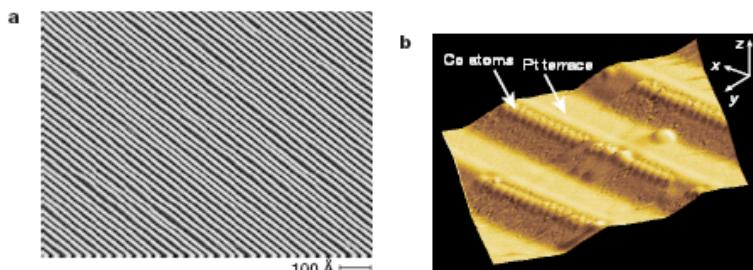
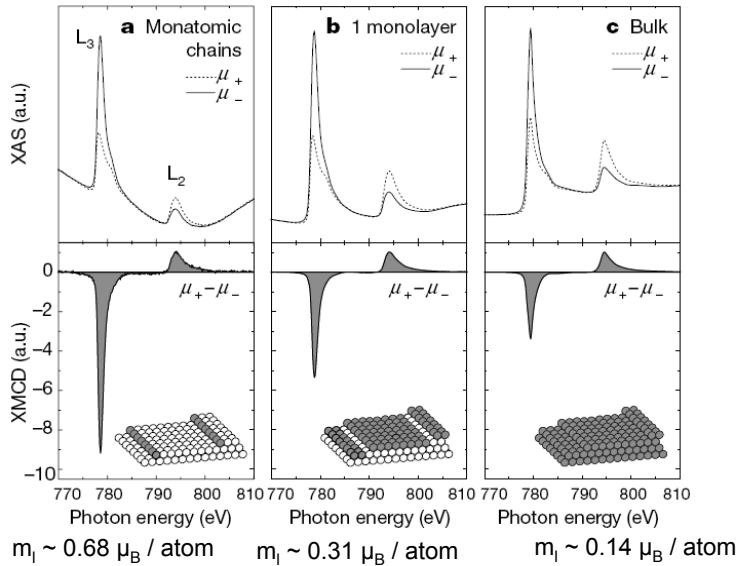


Figure 1 STM topographs of the Pt(997) surface. **a**, Periodic step structure (each white line represents a single step). The surface has a 6.45° miscut angle relative to the (111) direction; repulsive step interactions result in a narrow terrace width distribution centred at 20.2 Å with 2.9 Å standard deviation. **b**, Co monatomic chains decorating the Pt step edges (the vertical dimension is enhanced for better contrast). The monatomic chains are obtained by evaporating 0.13 monolayers of Co onto the substrate held at $T = 260\text{ K}$ and previously cleaned by ion sputtering and annealing cycles in ultrahigh vacuum (UHV). The chains are linearly aligned and have a spacing equal to the terrace width.

P. Gambardella et al. Nature 416, 301 (2002)

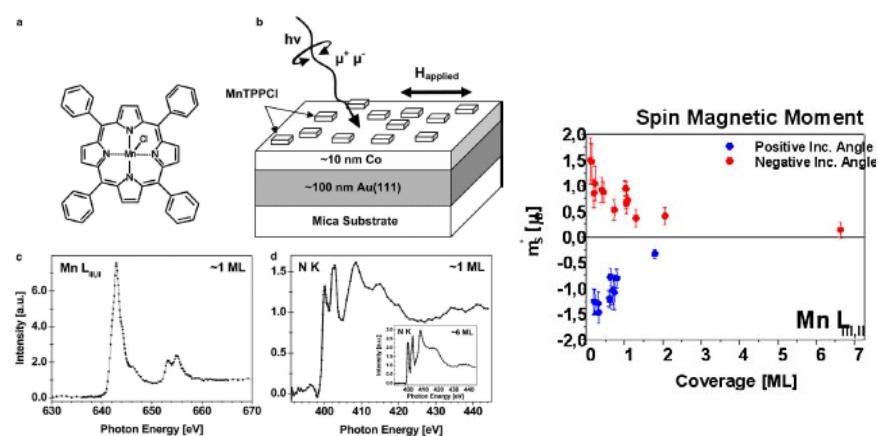
Increased orbital moment in Co chains

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P. Gambardella et al. Nature 416, 301 (2002)

Induced magnetic ordering in a molecular monolayer



A. Scheybal, T. Ramsvik, R. Bertschinger, M. Putero, F. Nolting, and T.A. Jung, Chemical Physics Letters 411, 214 (2005)

Outline

X-ray absorption spectroscopy (XAS)

Absorption process

Total electron yield mode

Examples

X-ray Magnetic Circular Dichroism (XMCD)

Basics

Example: Magnetocrystalline Anisotropy

Closer look at the absorption process

Multiplet effects

Example: Interface effect in Exchange Bias system

X-ray Magnetic Linear Dichroism (XMLD)

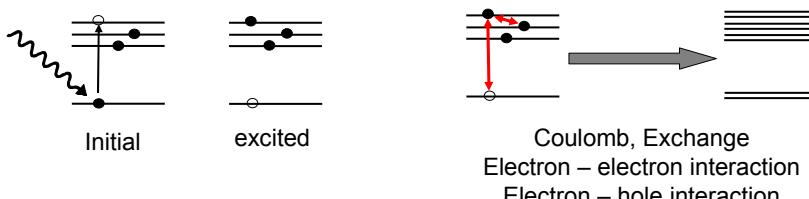
Basics

Combine it all

Summary Example: Laser control of an exchange bias system

Interactions

Single electron

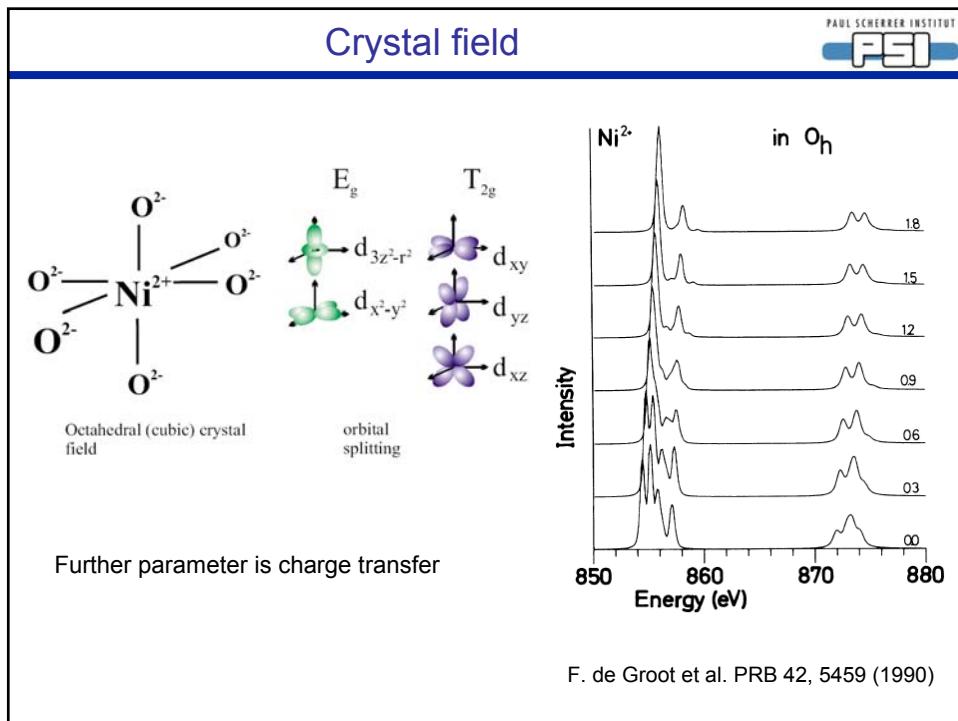
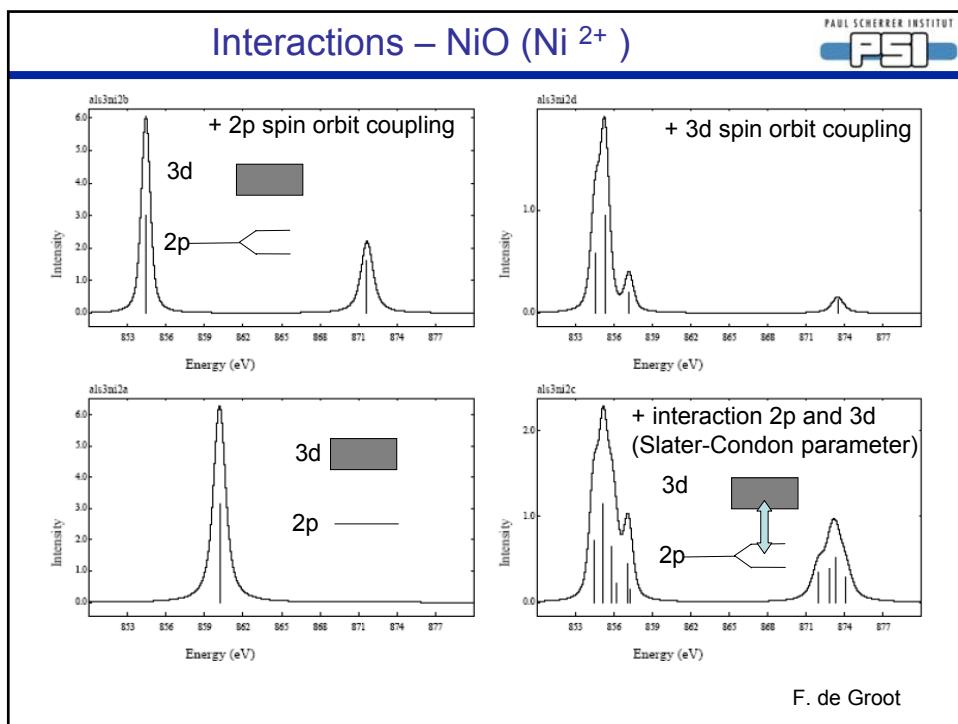


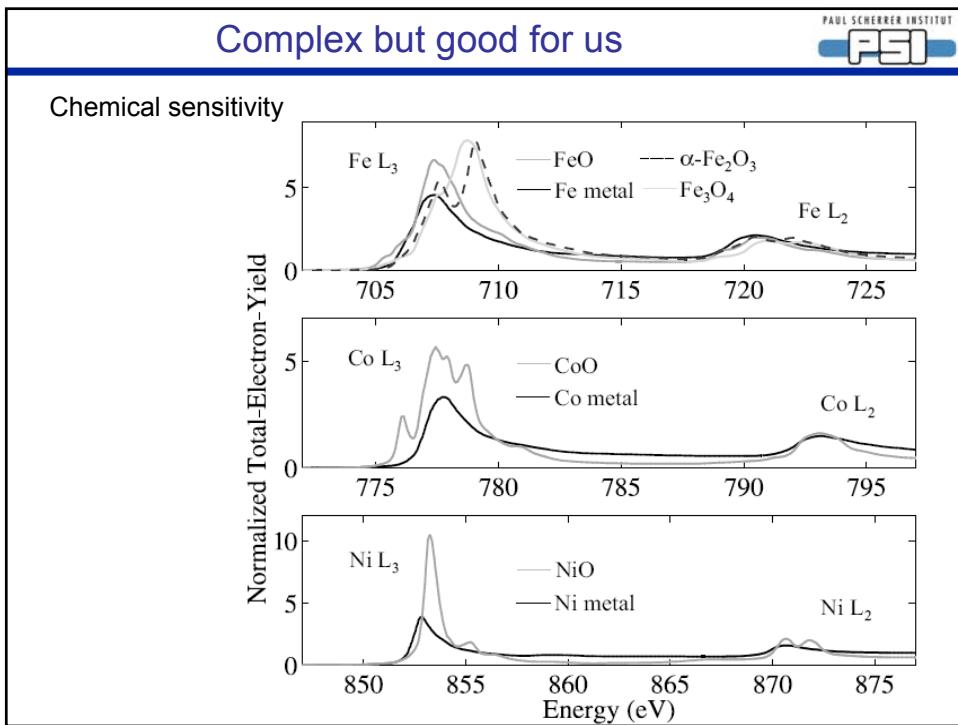
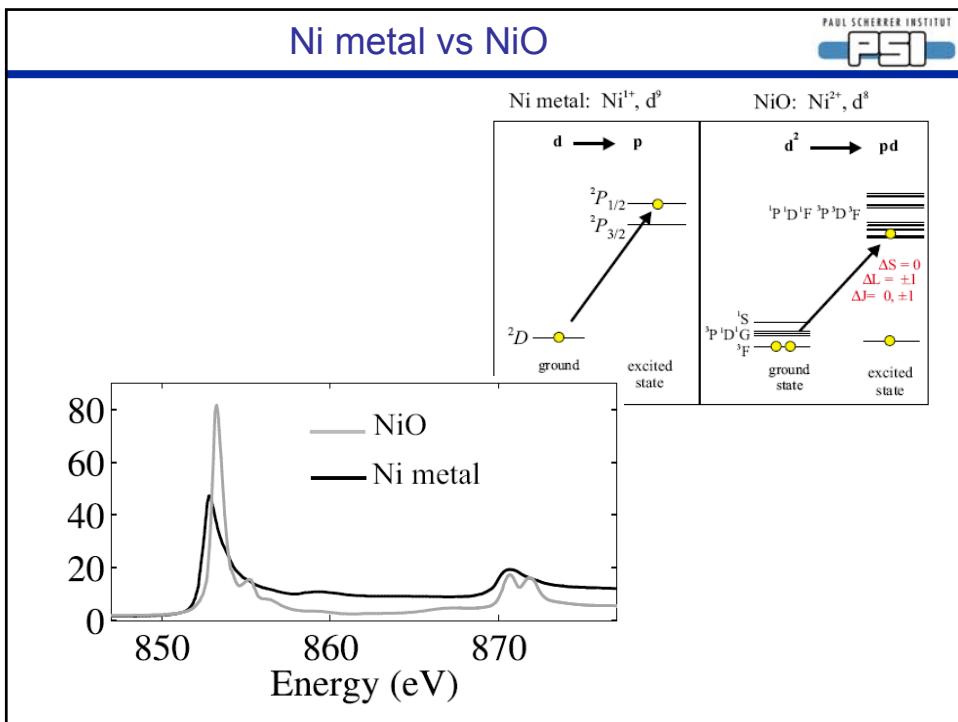
Coulomb, Exchange
Electron – electron interaction
Electron – hole interaction

Valence – Valence interaction : many body effects

Valence – Core interaction : multiplet effects

Hybridization between ground state and final state
leads to a multiplet structure of the spectrum





Exchange Bias

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Unidirectional anisotropy in FM adjacent to AFM
discovered 1956 by Meiklejohn and Bean

Effect of exchange bias is still poorly understood

Ideal Interface Model	Spin Flop Model	Domain Size Model
		<p>Uncompensated spins at AF domain boundaries</p>

Reviews: A.E. Berkowitz and K. Takano, J. Magn. Magn. Mater. 200(1-3), 552 (1999).
J. Nogues and I.K. Schuller, J. Magn. Magn. Mater. 192, 203 (1999)

Exchange biased Co/NiO multilayer

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Electron Yield (arb. units.)

Ni L₂-edge: Plot of Electron Yield vs Photon Energy (eV) from 868 to 874 eV. Peaks A and B are shown for Ni, NiO, and Co/NiO samples. A model curve is also shown.

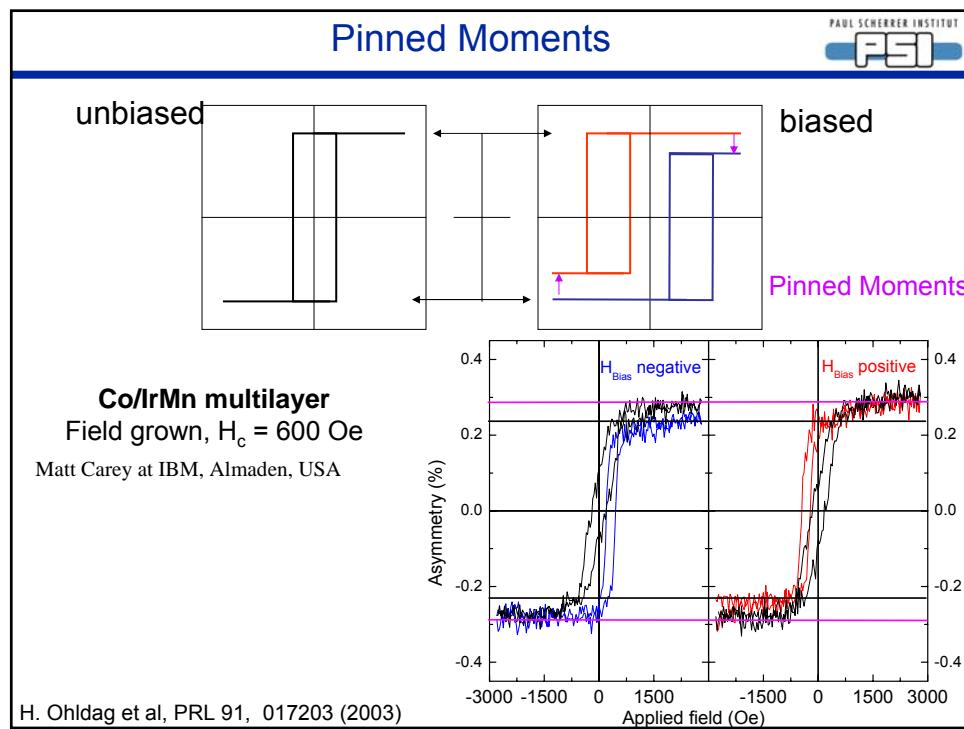
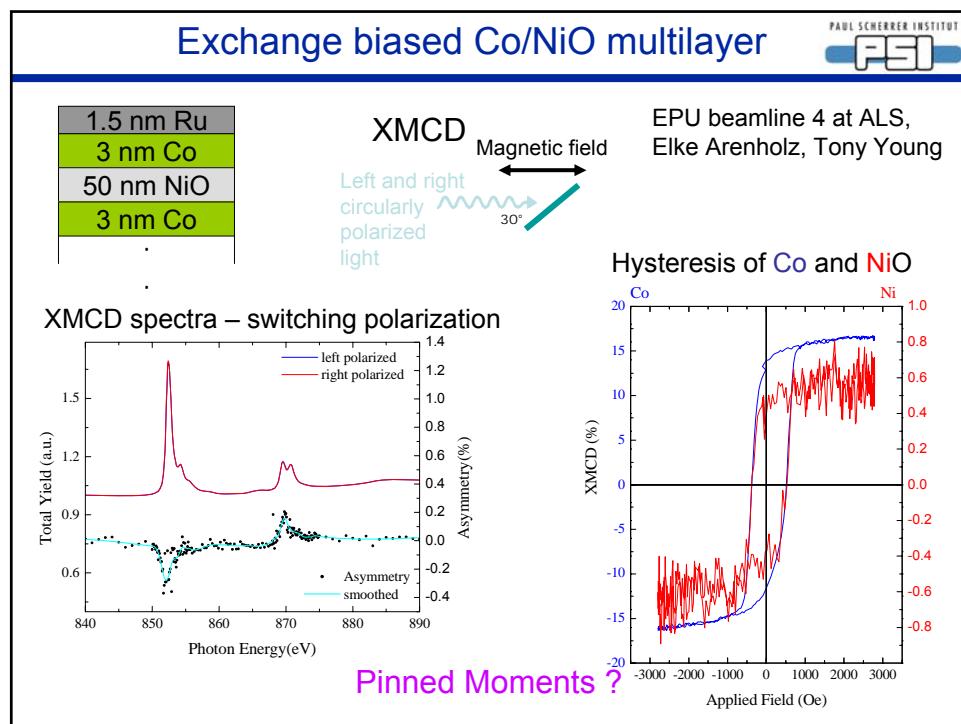
Co L₂-edge: Plot of Electron Yield vs Photon Energy (eV) from 776 to 780 eV. Peaks are shown for Co, CoO, and Co/NiO samples. A model curve is also shown.

Intensity thickness 2*l*(m) (%)

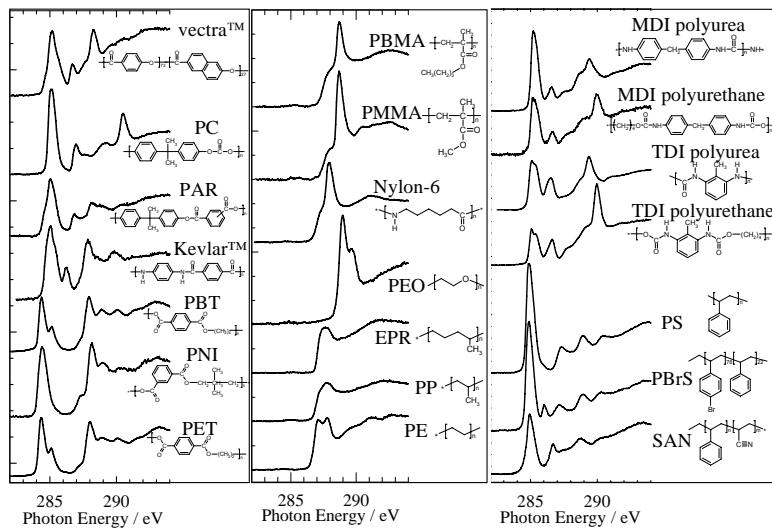
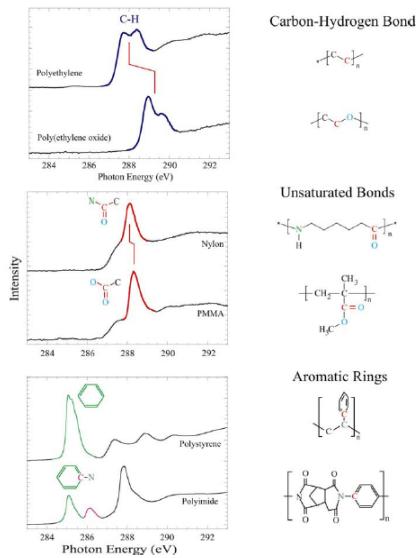
Plot of Intensity thickness 2*l*(m) (%) vs Annealing time at 600K (mins). The signal increases with annealing time, reaching a plateau around 1200 minutes. An inset shows the XRD pattern of the annealed sample.

Oxidation/reduction at the interface is responsible for increased coercivity

Tom Regan et al, Phys. Rev. B 64, 214422 (2001)



NEXAFS spectra of polymers: building block picture



Harald Ade (1997)

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Closer look at the absorption process

Multiplet effects

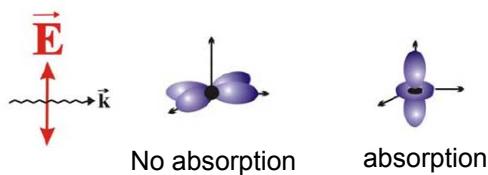
Example: Interface effect in Exchange Bias system

X-ray Magnetic Linear Dichroism (XMLD)

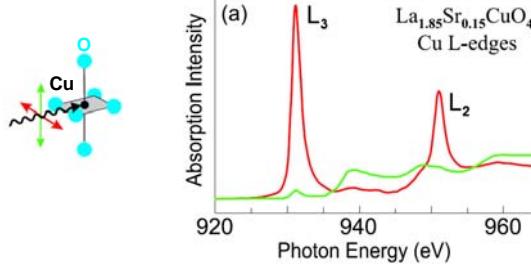
Basics

Interaction with linear light - charge

Excitation into 3d band

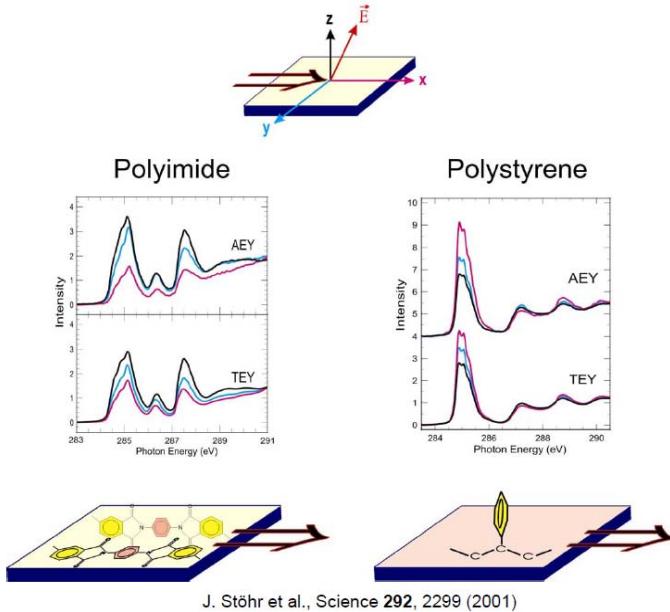


X-ray Natural linear dichroism
“search light effect”



C.T. Chen et al PRL 68, 2543 (1998)

Polarization Dependent NEXAFS Probes Bond Anisotropy at Surface

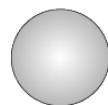


J. Stöhr et al., Science 292, 2299 (2001)

Interaction with linear light - magnetic

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Paramagnetic State



Electron charge density is isotropic
no linear dichroism

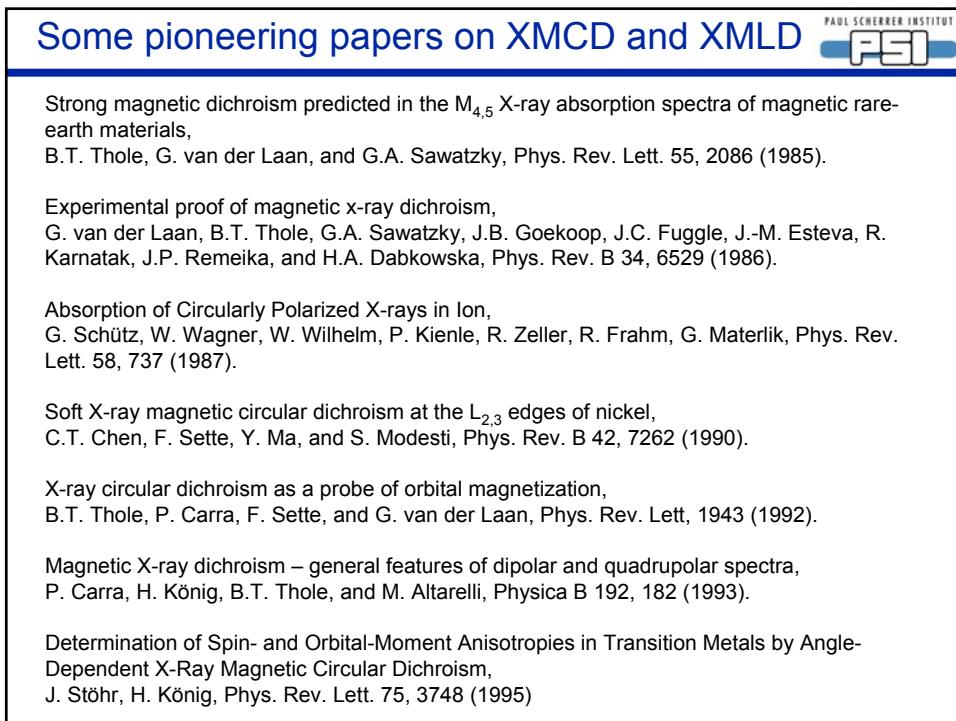
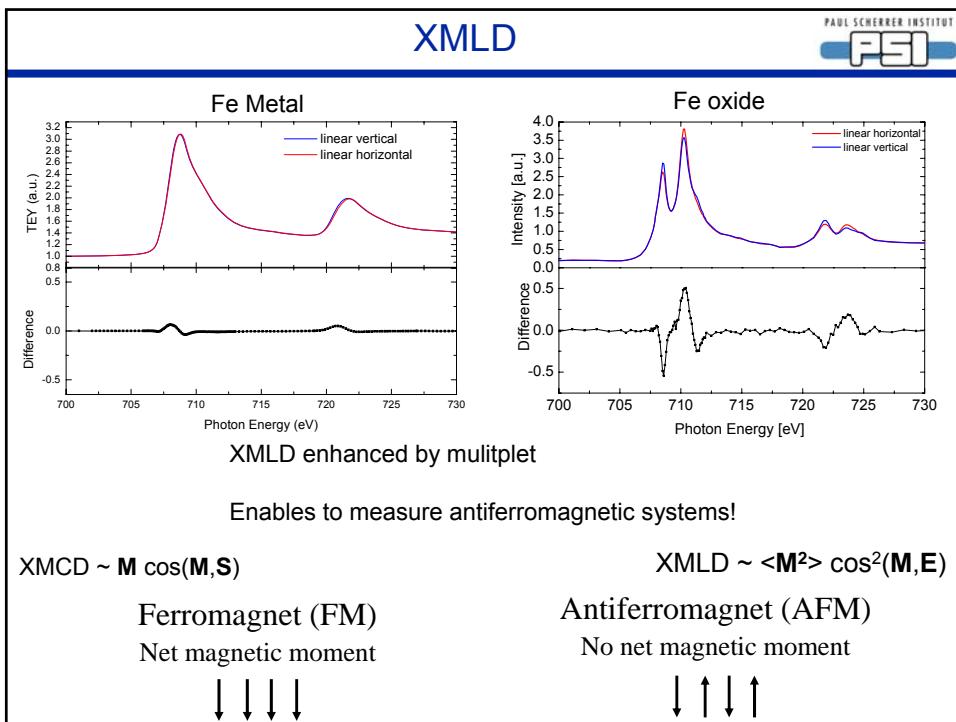
Aligned Magnetic State



Preferred spin axis
spin orbit coupling changes charge density
linear dichroism

XMD

X-ray Magnetic Linear Dichroism



Brief excursion
X-rays / Synchrotron generation



1895 Discovery of X-rays by Wilhelm Röntgen
1901 Nobel prize in physics

Image of hand of Albert von Kölliker
this is the second image, the first one, very
similar is said to be the hand of his wife

Wilhelm Röntgen

27. März 1845 in Lennep geboren.

1861 bis 1863 Technische Schule in Utrecht. Aus disziplinarischen Gründen, weil er irrtümlich für den Urheber einer Karikatur seines Klassenlehrers gehalten wurde, verwies man ihn ohne Abitur von der Schule.

1864 - 1868 Eidgenössischen Technischen Hochschule Zürich (ETH Zürich)
Maschinenbauingenieur

1869 promovierte Röntgen an der Universität Zürich in Physik mit „Studien über Gase“.

1870 begleitete er August Kundt als Assistent nach Würzburg.

1874 Habilitation Universität Straßburg die ihm die Universität Würzburg zuvor wegen seines fehlenden Abiturs verweigert hatte.

1875 außerordentlicher Professor für Physik und Mathematik an der Landwirtschaftlichen Akademie Hohenheim.

1876 eine Stelle als außerordentlicher Professor für Physik in Straßburg.

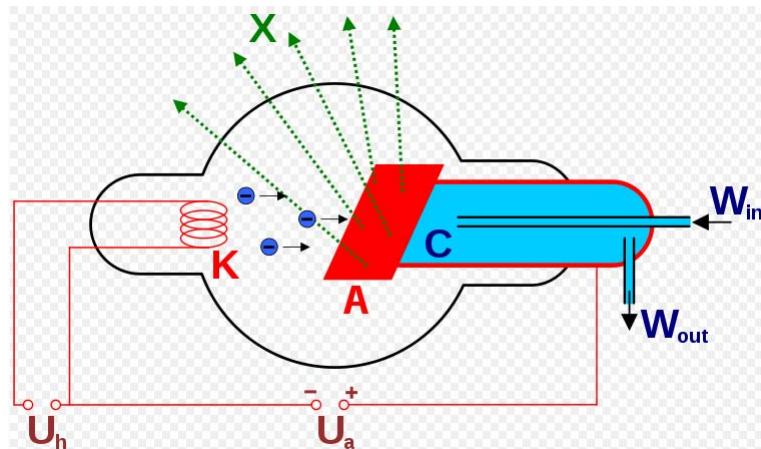
1879 ordentliche Professur in Gießen

1888 Professor der Experimentalphysik Würzburg.

1900 Professor an der Universität München

1923 verstorben

X-ray tube



Creation of electromagnetic radiation

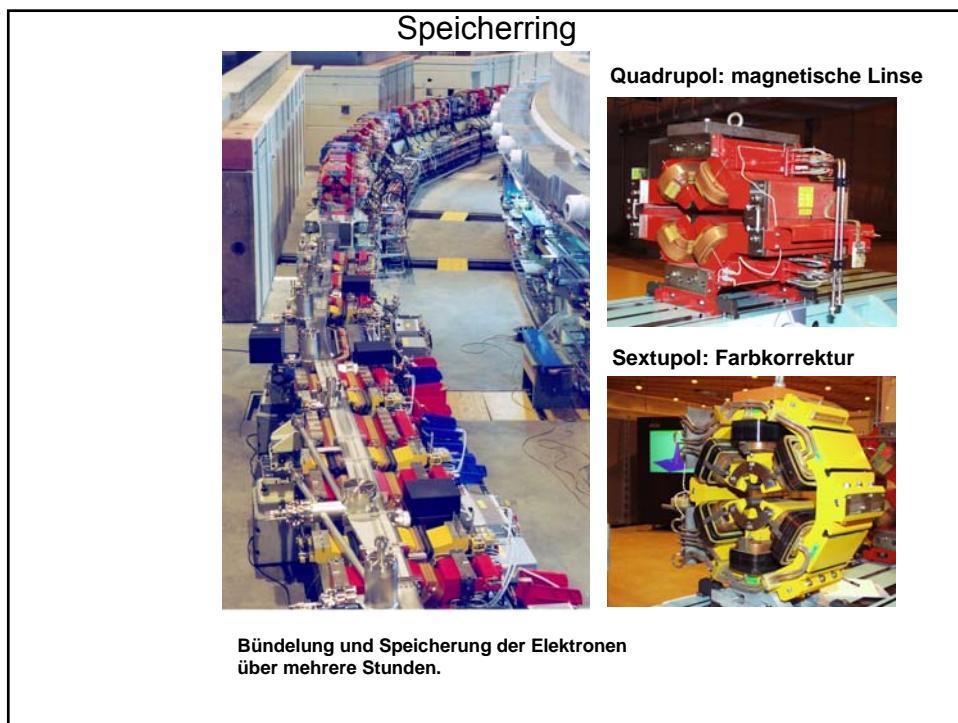
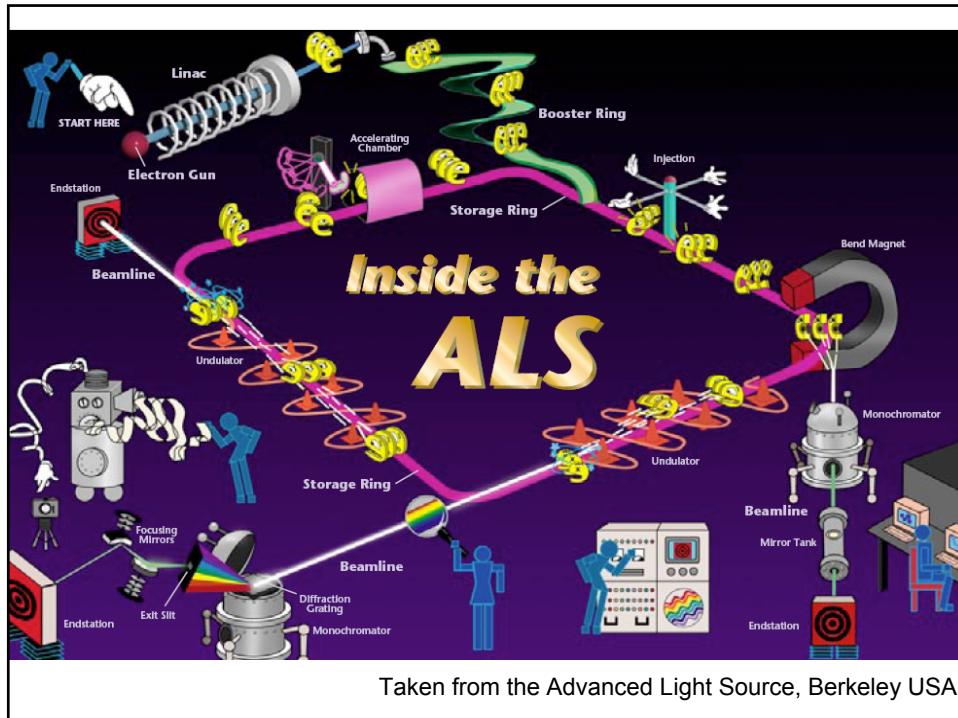
The Liénard–Wiechert field $\mathbf{E}(t)$ of a point charge q detected by an observer at a time t is determined by the distance r^* , the velocity v^* , and acceleration a^* of the charge at the emission or retarded time $t^* = t - r^*/c$. Defining $\beta^* = v^*/c$ we have

$$\begin{aligned} \mathbf{E}(t) = & \frac{q}{4\pi\epsilon_0} \underbrace{\frac{1 - (\beta^*)^2}{(r^*)^2 (1 - \mathbf{n}^* \cdot \beta^*)^3} [\mathbf{n}^* - \beta^*]}_{\text{velocity field}} \\ & + \frac{q}{4\pi\epsilon_0} \underbrace{\frac{1}{c^2 r^* (1 - \mathbf{n}^* \cdot \beta^*)^3} \{\mathbf{n}^* \times ([\mathbf{n}^* - \beta^*] \times \mathbf{a}^*)\}}_{\text{acceleration field}}. \quad (4.58) \end{aligned}$$

We have indicated all retarded quantities by an asterisk.



Swiss Light Source



polarized X-rays!

$$\mathbf{l} = m_e(\mathbf{r} \times \mathbf{v}) = m_e \mathbf{r}^2 \boldsymbol{\omega}$$

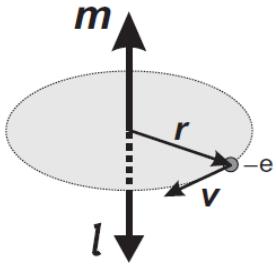
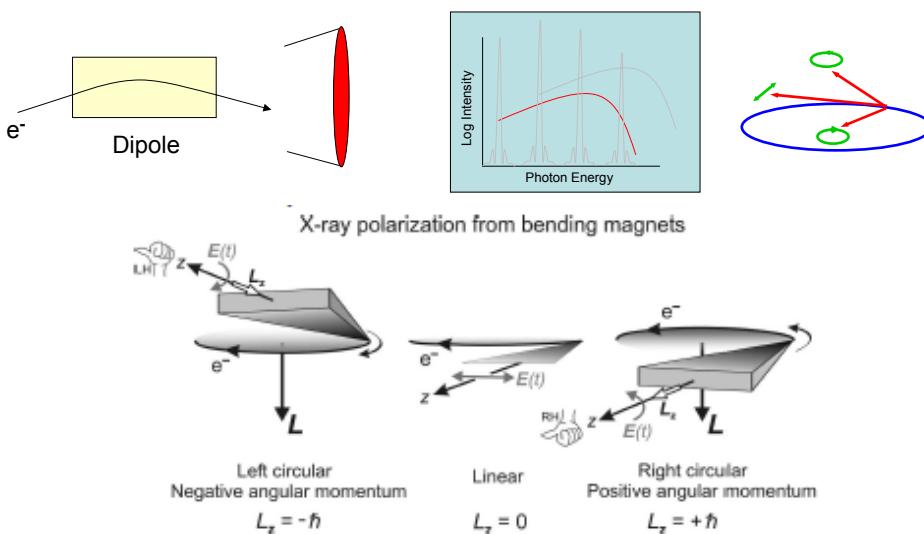
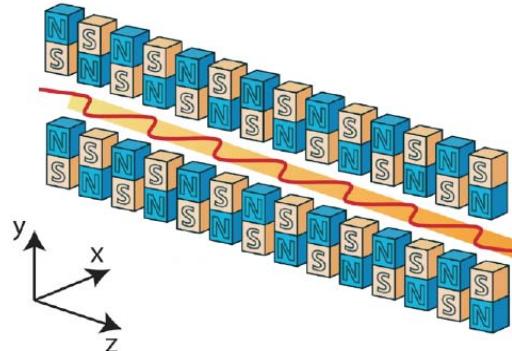
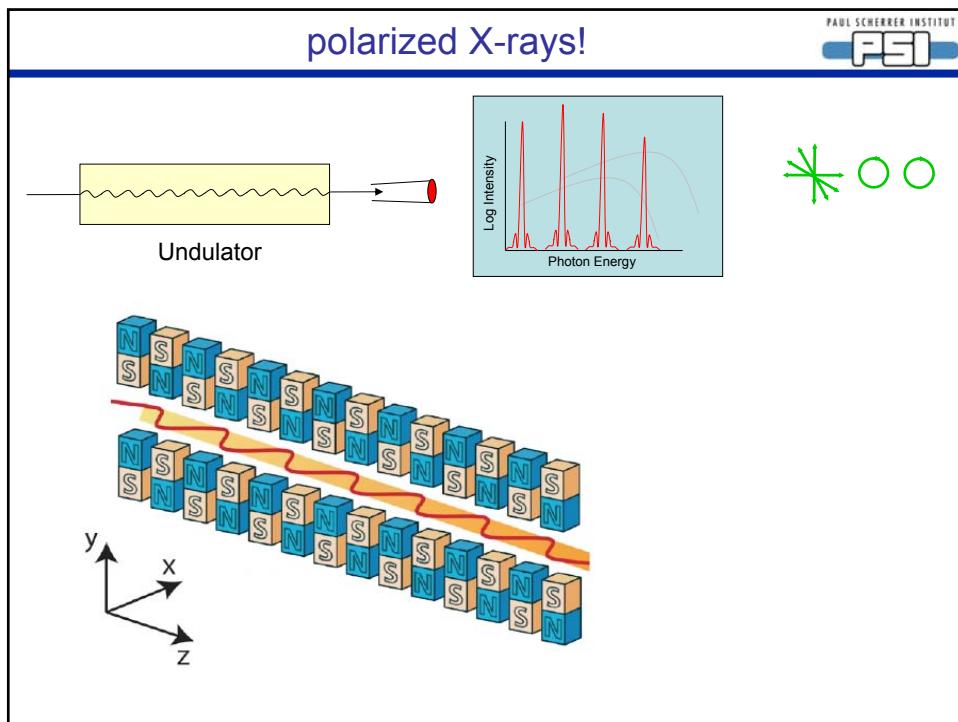


Fig. 3.5. Definition of the magnetic moment caused by an electron $q = -e$ that orbits around a center at a distance r with a tangential velocity v and angular velocity ω . The classical angular momentum \mathbf{l} of the rotating electron with mass m_e is also shown

polarized X-rays!

Slide Courtesy: Thomas Schmidt





Conclusions

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Polarization depend soft X-ray absorption spectroscopy is a powerful tool to study elemental resolved ferromagnetic and antiferromagnetic thin films and interfaces

Non-spatially resolved

A graph showing XMCD (%) on the y-axis (ranging from -20 to 20) versus Applied Field (Oe) on the x-axis (ranging from -3000 to 3000). Two curves are plotted: one for Cobalt (Co) and one for Nickel (Ni). Both curves show a sharp transition near 0 Oe, where the XMCD value changes from negative to positive values. The Ni curve has a higher peak than the Co curve.

spatially resolved ...next time

A spatial map showing XMCD values across a surface. The color scale ranges from -0.8 to 1.0, with red indicating positive values and purple indicating negative values. The map shows a complex, textured pattern of alternating positive and negative regions.

The technique

is sensitive to

- elemental composition
- chemical bonds
- structural parameters
- electronic structure
- magnetic properties

has a very broad range of application

- Magnetic multilayers
- Diluted systems
- Nanoparticles
- Molecular magnets
- ...