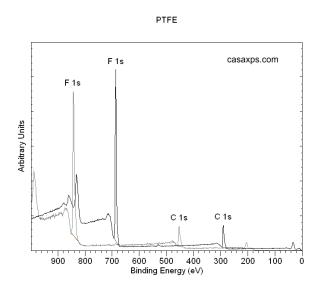
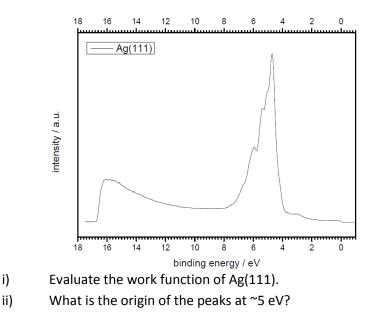
Exercise 1: XPS

- a) A MgK_α X-ray source is used to record the XPS spectrum of palladium. At which kinetic energies would one expect peaks? Tipp: The most prominent peaks will originate from the 3s, 3p and 3d shells.
- b) Look at the two spectra of Teflon on the left. What might cause the difference, given that it is the same sample? How could one prevent it?



Exercise 2: Ultraviolet photoelectron spectroscopy

- a) UPS is very similar to XPS. What are the main differences?
- b) The following spectrum was measured on a clean Ag(111) surface using UV-light from a He-I source.

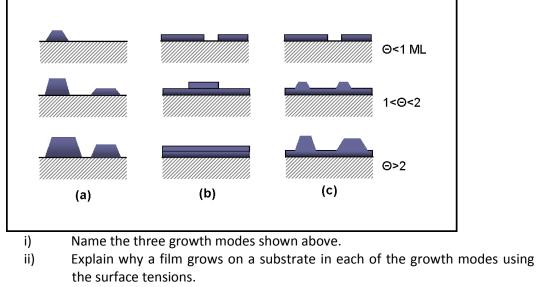


Exercise 3: Surface diffusion

- a) The hopping rate of a nitrogen atom on the Fe(100) surface is 10^{-3} s⁻¹ at 300 K and $3x10^{-2}$ s⁻¹ at 330 K. Estimate the diffusion coefficient and calculate the activation energy. Take into account that Fe is a bcc crystal with lattice parameter 2.87 Å. Assume the vibration frequency n_0 is $4.3x10^{12}$ Hz.
- b) Random-walk diffusion of Ag atoms occurs over the Si(111) v3xv3-Ag surface. Estimate the mean displacement of the atom in a time of 1 s and 1 h at 450 °C. $D_0 = 10^{-3} \text{ cm}^2 \text{s}^{-1}$, $E_{\text{diff}}=0.33 \text{ eV}$.

Exercise 4: Thin film growth

a) Thermodynamics



iii) How can you experimentally distinguish the three growth modes using XPS?

iv) How can you distinguish them with RHEED?

b) Kinetics: For the case of homoepitaxy, considering the parameters "substrate temperature" and "deposition rate": how should one choose those parameters in order to obtain 2D or 3D growth? Explain.